

LN52A650

52" Full High Definition TV with 1080p Resolution

SERIES **6** LCD TV



Samsung presents innovative HD design and performance on a brilliant 52-inch LCD screen. Samsung's exclusive Touch of Colour™ design accents the top and bottom edges of the bezel with a unique red tone. Full 1080p HD resolution and 50,000:1 dynamic contrast ratio offer stunning image detail and vivid colours. Auto Motion Plus 120Hz™ and Ultra Clear Panel technology work together to offer the smoothest motion and boldest bright and dark tones. Stay connected to the web via the InfoLink™ RSS feature. The Wiselink® USB 2.0 input and user-friendly interface gives you access to pictures and playlists from a specially designed Wheel Remote.

TV/Video

- Touch of Colour (ToC™) design
- InfoLink™ (RSS feeds of customizable stock quotes, weather and brief news articles)*
- Auto Motion Plus 120Hz™
- 1920 (H) x 1080 (V) HD pixel resolution
- 50,000:1 dynamic contrast ratio
- 18 bit video processor
- Wide color enhancer 2
- Fast 4ms response time
- NTSC/ATSC Tuners with Clear QAM
- Wheel Key Remote
- Ultra Clear Panel™
- Entertainment Mode
- Picture-in-Picture
- 1080p 24fps Real Movie

Audio

- Watts per channel: 10W x 2
- SRS TruSurround XT®
- Down-firing Speakers
- Auto volume leveller

Connections

- 4 HDMI™ — High Definition Multimedia Interface (3 rear/1 side)
- 2 Component Video Inputs (rear)
- S-Video Input (side)
- Composite Inputs (1 rear/1 side)
- PC Input (rear)
- RF Input (rear)
- Headphone Jack (side)
- Optical Sound Output (rear)
- Wiselink® (USB 2.0) (side)
- Ethernet 10/100 (rear)
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Samsung Ln52a650 Manual

Scott C. Dulebohn



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$$Cu + 2AgNO_3 \rightarrow Cu(NO_3)_2 + 2Ag$$

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you have 0.216232 grams mole 21.3 grams 0.66 moles 32 g m

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calculated mass of the product expected $\times 100$ in this reaction a yield 80 is anticipated

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g $\text{Al} \times 1 \text{ mol Al} \times 2 \text{ mol Al}_2\text{O}_3 \times 101.964 \text{ g Al}_2\text{O}_3 / 302.3 \text{ g Al}_2\text{O}_3 = 26.982 \text{ g Al}_2\text{O}_3$ 4 mol $\text{Al} \times 1 \text{ mol Al}_2\text{O}_3$ c determine the percent yield

yield actual $\times 100$ $26.0 \text{ g} \times 100 / 86.01 \text{ theoretical} = 302.3 \text{ g}$

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C is the limiting reagent 4 33 g of H_2 are left over 13

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