Quantum Numbers

- •
- Used to _____ an ____ in an _____

42

- Represents _____ energy level of ______

______# of ______ in an

Example: What is the maximum number of electrons that can be in the _____ main energy level?

1

- The
- Describes the ______ within an ______
- of orbital _____possible in ______

Orbital Shapes

designated

- level 1: _____
- fevel 2: _______
- level 3:
- level 4: ______

How many electrons can each sublevel hold?

s = 1 orbital x 2 e /orbital = _____e

p = 3 orbitals × 2 e /orbital = ______e

d = 5 orbitals x 2 e /orbital = _____e

f = 7 orbitals x 2 e /orbital = ____e

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