

Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms present in a sample
unit.

Atomic

^{12}C mass 12 atomic mass

^{16}O mass 16 atomic mass
unit

Molecular

$\text{C}_{10}\text{H}_{12}\text{O}_4$ atomic mass

$\text{C}_{10}\text{H}_{12}\text{O}_4$ mass 208 atomic mass

$\text{C}_5\text{H}_6\text{O}_2$ atomic mass

Note Taking Guide Episode 601 Chemical Formula

N Colangelo



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