

FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

Homologous series	Structure of functional group	Example of a compound	
		Name	Structural formula
alkanes	Only C-H and C-C single bonds	ethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
alkenes	$\begin{array}{c} \diagup & & \diagdown \\ & \text{C}=\text{C} & \\ \diagdown & & \diagup \end{array}$	ethene	$\begin{array}{c} \text{H} & & \text{H} \\ & \diagdown & \diagup \\ & \text{C}=\text{C} & \\ & \diagup & \diagdown \\ \text{H} & & \text{H} \end{array}$
alkynes	$-\text{C}\equiv\text{C}-$	ethyne	$\text{H}-\text{C}\equiv\text{C}-\text{H}$
haloalkanes (alkyl halides)	$\begin{array}{c} \\ -\text{C}-\text{X} \\ \\ \text{(X = F, Cl, Br, I)} \end{array}$	bromoethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{Br} \\ & & \\ \text{H} & & \text{H} \end{array}$
alcohols (alkanols)	$\begin{array}{c} \\ -\text{C}-\text{O}-\text{H} \\ \end{array}$	ethanol	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
aldehydes	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{H} \end{array}$	ethanal	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ \\ \text{H} \end{array}$
ketones	$\begin{array}{c} \text{O} \\ \\ -\text{C}- \\ \end{array}$	propan-2-one	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C} & - & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$
carboxylic acids	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{H} \end{array}$	ethanoic acid	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ \\ \text{H} \end{array}$
esters	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{C}- \\ \end{array}$	methyl ethanoate	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}- & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon chain.

prefix

What and where are the substituents?

parent

How many carbons?

suffix

What is the functional group/homologous series?

di - 2

tri - 3

tetra - 4

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct



prefix, suffix and parent!

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Tetra-chloro-metha

Physical Science Organic Compounds Study Guide

Jiben Roy



Physical Science Organic Compounds Study Guide:

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