

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

12.01

Na

Sodium

14.01

Mg

Magnesium

19.00

K

Potassium

20.02

Ca

Calcium

21.03

Sc

Scandium

22.01

Ti

Titanium

23.02

V

Vanadium

24.06

Cr

Chromium

25.04

Mn

Manganese

26.09

Fe

Iron

27.01

Co

Cobalt

28.01

Ni

Nickel

29.08

Cu

Copper

30.07

Zn

Zinc

31.04

Ga

Gallium

32.06

Ge

Germanium

33.05

As

Arsenic

34.09

Se

Selenium

35.45

Br

Bromine

36.97

Kr

Krypton

37.40

Rb

Rubidium

38.08

Sr

Strontium

39.10

Y

Yttrium

40.08

Zr

Zirconium

41.02

Nb

Niobium

42.01

Mo

Molybdenum

43.02

Tc

Technetium

44.06

Ru

Ruthenium

45.07

Rh

Rhodium

46.02

Pd

Palladium

47.03

Ag

Silver

48.02

Cd

Cadmium

49.09

In

Indium

50.12

Sn

Tin

51.00

Sb

Antimony

52.04

Te

Tellurium

53.09

I

Iodine

54.09

Xe

Xenon

55.08

Cs

Cesium

56.01

Ba

Barium

57.25

La

Lanthanum

58.93

Ce

Cerium

59.91

Pr

Praseodymium

60.92

Nd

Neodymium

61.91

Pm

Promethium

62.93

Sm

Samarium

63.95

Eu

Europium

64.93

Gd

Gadolinium

65.93

Tb

Terbium

66.93

Dy

Dysprosium

67.93

Ho

Holmium

68.93

Er

Erbium

69.92

Tm

Thulium

70.91

Yb

Ytterbium

71.90

Lu

Lutetium

72.91

Hf

Hafnium

73.92

Ta

Tantalum

74.92

W

Tungsten

75.94

Re

Rhenium

76.94

Os

Osmium

77.94

Ir

Iridium

78.96

Pt

Platinum

79.90

Au

Gold

80.91

Hg

Mercury

81.90

Tl

Thallium

82.91

Pb

Lead

83.90

Bi

Bismuth

84.96

Po

Polonium

85.91

At

Astatine

86.91

Rn

Radon

87.63

Fr

Francium

88.10

Ra

Radium

89.00

Ac

Actinium

90.02

Th

Thorium

91.02

Pa

Protactinium

92.02

U

Uranium

93.03

Np

Neptunium

94.04

Pu

Plutonium

95.04

Am

Americium

96.06

Cm

Curium

97.06

Bk

Berkelium

98.07

Cf

Californium

99.07

Es

Einsteinium

100.08

Fm

Fermium

101.08

Md

Mendelevium

102.08

No

Nobelium

103.08

Lr

Lawrencium

101

Si

Silicon

atomic number

atomic weight

symbol

black solid

blue liquid

red gas

white synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

LABS

5.09 B Boron

6.01 C Carbon

7.01 N Nitrogen

8.01 O Oxygen

9.01 F Fluorine

10.01 Ne Neon

13.01 Al Aluminum

14.01 Si Silicon

15.01 P Phosphorus

16.01 S Sulfur

17.01 Cl Chlorine

18.01 Ar Argon

31.01 Ga Gallium

32.01 Ge Germanium

33.01 As Arsenic

34.01 Se Selenium

35.01 Br Bromine

36.01 Kr Krypton

49.01 In Indium

50.01 Sn Tin

51.01 Sb Antimony

52.01 Te Tellurium

53.01 I Iodine

54.01 Xe Xenon

81.01 Tl Thallium

82.01 Pb Lead

83.01 Bi Bismuth

84.01 Po Polonium

85.01 At Astatine

86.01 Rn Radon

(113)

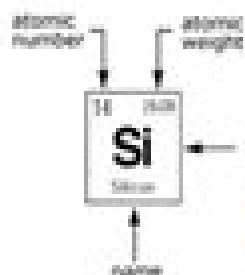
(114)

(115)

(116)

(117)

(118)



black solid
blue liquid
red gas
white synthetically prepared
over 1000 degree

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

Physical Science Lab Periodic Table

Michael Hoelscher



Physical Science Lab Periodic Table:

Science Action Labs Physical Science (eBook) Edward Shevick, 1998-03-01 Matter and Motion These easy to use hands on explorations are just what you need to get your science curriculum and your students into action The Periodic Table Edward Shevick, 2008-09-01 Implement Newton's First Law of Motion as a teaching principle with this packet students bodies at rest need many hands on activities impressed forces to learn compelling change This collection of Physical Science Action Labs will give your students plenty of experience with matter The labs include determining characteristics of matter focusing specifically on the periodic table *Exploring Physical Science in the Laboratory* John T. Salinas, 2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging *Exploring Physical Science in the Laboratory* guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Physics I For Dummies Steven Holzner, 2016-05-17 The fun and easy way to get up to speed on the basic concepts of physics For high school and undergraduate students alike physics classes are recommended or required courses for a wide variety of majors and continue to be a challenging and often confusing course *Physics I For Dummies* tracks specifically to an introductory course and keeping with the traditionally easy to follow Dummies style teaches you the basic principles and formulas in a clear and concise manner proving that you don't have to be Einstein to understand physics Explains the basic principles in a simple clear and entertaining fashion New edition includes updated examples and explanations as well as the newest discoveries in the field Contains the newest teaching techniques If just thinking about the laws of physics makes your head spin this hands on friendly guide gets you out of the black hole and sheds light on this often intimidating subject **Science Mysteries Explained** Anthony Fordham, 2014-06-05 Ideal for the armchair science enthusiast *Idiot's Guides Science Mysteries Explained* takes a question answer based approach to teach readers a wide variety of topics in Earth Science Life Science Chemistry Physics and Cosmology Using helpful full color illustrations and expert information this book features 130 fascinating questions and answers to satisfy any scientist wannabe

Proceedings of the High School Conference of ... Horace Adelbert Hollister, 1928 **Proceedings of the High**

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