

Grade 8 Science Notes: Chemical Reactions

Chemical Reactions are processes where substances are transformed into new substances with different properties.

Types of Chemical Reactions:

1. Combination Reaction:

- Two or more substances combine to form one product.
- Example: Hydrogen + Oxygen $\rightarrow$ Water.

2. Decomposition Reaction:

- A compound breaks down into simpler substances.
- Example: Water $\rightarrow$ Hydrogen + Oxygen.

3. Displacement Reaction:

- An element replaces another in a compound.
- Example: Zinc + Copper Sulfate $\rightarrow$ Zinc Sulfate + Copper.

4. Combustion Reaction:

- A substance reacts with oxygen, releasing energy.
- Example: Methane + Oxygen $\rightarrow$ Carbon Dioxide + Water + Energy.

Reactants and Products:

- Reactants: Starting substances.
- Products: New substances formed.

Indicators of a Chemical Reaction:

1. Color Change.
2. Gas Production (bubbles).
3. Temperature Change (heat release/absorption).
4. Formation of Precipitate.

Importance:

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SJ Ball



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The Enigmatic Realm of **Physical Science Grade8 Chemical Reaction Pearson**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Physical Science Grade8 Chemical Reaction Pearson** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those who partake in its reading experience.

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