

2.1**PROPERTIES OF MATTER****Section Review****Objectives**

- Identify physical properties and physical changes
- Distinguish intensive properties from extensive properties
- Differentiate among three states of matter

Vocabulary

- | | | |
|----------------------|---------------------|-------------------|
| • mass | • substance | • gas |
| • volume | • physical property | • vapor |
| • extensive property | • solid | • physical change |
| • intensive property | • liquid | |

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Properties used to describe matter can be classified as _____ 1 _____ or _____ 2 _____. The _____ 3 _____ of an object is a measure of the amount of matter the object contains. The _____ 4 _____ of an object is a measure of the space occupied by the object. An extensive property is one that depends on the _____ 5 _____ of matter. An intensive property is one that depends on the _____ 6 _____ of matter.

A _____ 7 _____ is matter that has uniform and definite composition. A solid has a definite _____ 8 _____ and _____ 9 _____. A liquid has a definite volume, but takes the _____ 10 _____ of its container. A _____ 11 _____ takes both the shape and volume of its container.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____
11. _____

Part B True-False

Classify each of these statements as always true, *AT*; sometimes true, *ST*; or never true, *NT*.

- _____ 11. Matter has mass and occupies space.
- _____ 12. A liquid has a definite shape.
- _____ 13. Heating a solid to 200°C will cause it to change to a liquid.
- _____ 14. Gases are easier to compress than liquids.

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Peipei Pang



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