

_____ 11) Which one of the following processes results in an increase in the atomic number?

- A) alpha decay B) gamma decay C) fission D) positron decay E) beta decay

_____ 12) Of the following processes, which one changes the atomic number?

- A) beta decay B) positron decay C) electron capture D) alpha decay E) All

_____ 13) Which type of radioactive decay results in no change in mass number and atomic number for the starting nucleus?

- A) beta B) positron C) alpha D) electron capture E) gamma

_____ 14) Alpha decay produces a new nucleus whose _____ than those respectively of the original nucleus.

- A) atomic number is 2 more and mass number is 4 more
B) atomic number is 2 less and mass number is 2 less
C) atomic number is 1 less and mass number is 2 less
D) atomic number is 2 less and mass number is 4 less
E) atomic number is 2 more and mass number is 2 less

_____ 15) What happens to the mass number and the atomic number of an element when it emits gamma radiation?

- A) The mass number increases by four and the atomic number increases by two.
B) The mass number and atomic numbers remain unchanged.
C) The mass number decreases by four and the atomic number decreases by two.
D) The mass number remains unchanged while the atomic number decreases by one.
E) The mass number remains unchanged while the atomic number increases by one.

Calculations - For the following problems, show all work.

16) Write the nuclear equation for the positron decay of $^{24}_{14}\text{Si}$

17) What is the remaining atom after the beta decay of K-42?

18) Radium undergoes alpha decay. The product of this reaction also undergoes alpha decay. What is the product of this second decay reaction?

19) Bombardment of uranium-235 with a neutron generates tellurium-135, 3 neutrons, and what other product? Show your work.

Physical Science Chapter 10 Nuclear Chemistry

Wordwise Answer Key

Sabine Zange



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