EASEL ACID AND BASE QUIZ

Multiple Choice: Select the most con-	rect answer. (15 K)	
1. Which of the following indicates a	n acidic solution?	
A) $pH = 7.5$	B) eH = 7	C) pH = 6.5
D) more than one correct response		and have
2. Which of the following statements	about weak acids is correct	2
A) Weak acids always contain C aton	this.	
B) A percentage dissociation of 50%		
C) Weak acid molecules readily lose	their acidic hydrogen atoms	
D) more than one correct response		
E) none of the above		
3. Which of the following equations :	removements on acid-base mount	Senitración reaction?
	B) HNO: + KOH -	
C) Ba(OH), + Na ₂ SO ₄ → BaSO ₄ + 22	NaOH D) more than one co	rrect response
E) none of the above		52,500,000 , 50,000,00
4. If the pH of a solution increases fro	on 4 0 to 5 0 the hadronius	s been reconstructions
	B) increases by a factor of 1	
	D) decreases by a factor of	
C) becreases by a lactor of 1.3.	D) the creases by a factor of	LOOL
S. The most basic oxides are formed	from elements found in the	region of the
periodic table.		
A) right B) left C) center D) to	op E) none of the above	
6. When CaO(s) reacts with water wh	sich of the following is true?	
a) The solution contains OH and o	Ca ^{1*} _(m) b) The solu	ation contains O2 (as) and Ca2 (as)
c) The solution contains O2 to OH to	o, and Ca ²⁴ (sq) d) The solut	ion contains CaO _(aq)
7. A weak hase:		
A) partially dissociates in water to give	ve a little OFF out	
B) fully reacts with water to form OH	Circle	
C) completely dissociates to form OF	T _{OND}	
D) has a very high pH		
E) contains a group 1 metal and a hye	firovide	
8. Which of the following is NOT a p	property of bases?	
	B) They taste bimer and fee	slippery
C) They have a pH greater than 7.	D) They turn red litmus blu	e E) They neutralize acids.

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