

Use the following Standard Reduction Potentials below to answer questions 23-24.

Half-reaction	$E^\circ$
$\text{Ag}^+ + \text{e}^- \rightarrow \text{Ag (s)}$	0.80 V
$\text{Co}^{2+} + 2\text{e}^- \rightarrow \text{Co (s)}$	-0.28 V

23. What is the standard cell potential,  $E^\circ_{\text{cell}}$  for the Ag-Co cell.

a. 0.52 V

b. 0.66 V

c. 1.08 V

d. 1.88 V

24. The correct redox reaction for the Ag-Co cell is which of the following:



# Reduction Potential Answer Key

**Oswaal Editorial Board**



## **Reduction Potential Answer Key:**

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