

## Practice Problems: Redox Reactions (Answer Key)

1. Determine the oxidation number of the elements in each of the following compounds:



**H: +1, O: -2, C: +4**



**N: 0**



**Zn: 2+, H: +1, O: -2**



**N: +3, O: -2**



**Li: +1, H: -1**



**Fe: +8/3, O: -2**

2. Identify the species being oxidized and reduced in each of the following reactions:



**Cr<sup>+</sup>: oxidized, Sn<sup>4+</sup>: reduced**



**Hg<sup>2+</sup>: reduced, Fe: oxidized**



**As: oxidized, Cl<sub>2</sub>: reduced**

3. Would you use an oxidizing agent or reducing agent in order for the following reactions to occur?

# Redox Reaction Problems With Answer Key

**M Lipman**



## **Redox Reaction Problems With Answer Key:**

## Whispering the Secrets of Language: An Psychological Journey through **Redox Reaction Problems With Answer Key**

In a digitally-driven world where displays reign great and immediate connection drowns out the subtleties of language, the profound secrets and psychological subtleties hidden within phrases usually move unheard. However, situated within the pages of **Redox Reaction Problems With Answer Key** a captivating literary prize blinking with natural feelings, lies a fantastic journey waiting to be undertaken. Written by an experienced wordsmith, that wonderful opus invites visitors on an introspective journey, gently unraveling the veiled truths and profound impact resonating within the very material of each and every word. Within the mental depths of the poignant evaluation, we will embark upon a heartfelt exploration of the book is key themes, dissect their charming publishing style, and yield to the powerful resonance it evokes strong within the recesses of readers hearts.

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