#### Electrochemistry Class 12 Important Questions Very Short Answer Type

Question 1.

What is meant by 'limiting molar conductivity'? (All India 2010)

Answer:

The molar conductivity of a solution at infinite dilution is called limiting molar conductivity and is represented by the symbol  $\Lambda_{-}$ .

Question 2.

Express the relation between conductivity and molar conductivity of a solution held in a cell. (Delhi 2011)

Answer:

 $\Lambda_m = KC = Conductivity Concentration$ 

Question 3.

What is the effect of catalyst on:

(i) Gibbs energy (ΔG) and

(ii) activation energy of a reaction? (Delhi 2017)

Answer:

(i) There will be no effect of catalyst on Gibbs .energy.

(ii) The catalyst provides an alternative pathway by decreasing the activation energy of a reaction.

Question 4.

What is the effect of adding a catalyst on

(a) Activation energy (Ea), and

(b) Gibbs energy (AG) of a reaction? (All India 2017)

Answer:

(a) On adding catalyst in a reaction, the activation energy reduces and rate of reaction is fastened.

(b) A catalyst does not alter Gibbs energy (AG) of a reaction.

# Electrochemistry Class 12 Important Questions Short Answer Type -I [SA – I]

Question 5.

Two half cell reactions of an electrochemical cell are given below:

MnO·.(aq) + 8H· (aq) + 5e → Mn<sup>-</sup> (aq) + 4H<sub>1</sub>O (I), E\* = + 1.51 V

 $Sn^{-}(aq) \rightarrow 4 Sn^{-}(aq) + 2e^{-}, E^{+} = + 0.15 V$ 

Construct the redox equation from the two half cell reactions and predict if this reaction favours formation of reactants or product shown in the equation. (All India 2009)

Answer:

The reactions can be represented at anode and at cathode in the following ways:

# **Redox And Electrochemistry Regents Short Answer Questions**

**J Spring** 

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