Chapter 12 - Stoichiometry

Note Taking Guide: Episode 801 Stoichiometry

-Study of the QUANTITY relationships in a CHEMICAL REACTION

-Based on BALANCED equations

 $2 \text{ Mg} + \text{O}_2$ \Rightarrow 2 MgO

1. Particles

2 atoms Mg 2 atoms O 2 atoms Mg 2 atoms O

2 atoms Mg 1 molecule O₂ 2 molecule MgO

10 atoms Mg 5 molecule O₂ 10 molecule MgO

2 (6.02 x 10²³) 1 (6.02 x 10²³) 2 (6.02 x 10²³) atoms Mg molecule O₂ molecule MgO

2. Moles

2 mol Mg 1 mol O₂ 2 mol MgO

The COEFFICIENTS in a BALANCED EQUATION give the MOLE RATIO for the SUBSTANCES involved in the REACTION.

Lab: Episode 801 - Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO₃) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g

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Note Taking Guide Episode 801 Stoichiometry:

Decoding Note Taking Guide Episode 801 Stoichiometry: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Note Taking Guide Episode 801 Stoichiometry**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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