

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ followed _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with elemental sulfur, aluminum chloride is produced



moles of Al

1	Al	3	S
2	Al	2	AlS ₃
3	Al	2	AlS ₃

2. If you start with 4 moles of Al, how many moles of AlS₃ will be produced?

2 moles Al (moles AlS₃) and 3 moles S (moles AlS₃)

Note Taking Guide Episode 801 Answers

Siddappa N.Byrareddy



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