

Blue cards participate _____ of the possible _____ of atoms
 1. _____
 2. _____

Atomic fraction number _____ of particles _____ in the particles _____ of an
 atom _____

Relative mass an set particles _____ \rightarrow an set particles _____ \rightarrow \rightarrow

Isotopes atoms _____ of an element _____ that have different _____ numbers
 of neutrons _____

Isotopes of Hydrogen

Hydrogen-1

1. _____ protons and _____ neutrons

Hydrogen-2

1. _____ protons and _____ neutrons

Hydrogen-3

1. _____ protons and _____ neutrons

Mass fraction mass _____ number of particles _____ and _____ in an
 atom _____

Example: Carbon-12 _____ 12.011 _____ 12.011 _____ 12.011 _____

Particle	Charge	Mass	Location
Proton	+1	1.6726 x 10 ⁻²⁷ kg	Nucleus
Neutron	Neutral	1.6749 x 10 ⁻²⁷ kg	Nucleus
Electron	-1	9.1091 x 10 ⁻³¹ kg	Electron cloud

Atomic mass average _____ of the masses _____ of all the isotopes of _____

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Note Taking Guide Episode 302 Answers

Jeff Veyera

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

Note Taking Guide Episode 302 Answers:

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