

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ followed _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



Write the balanced chemical equation for the reaction



2. If you start with 4 moles of Al , how many moles of $AlCl_3$ will be produced?

1. $2AlCl_3 + 3H_2O \rightarrow 2Al(OH)_3 + 3HCl$

Note Taking Guide Episode 80stoichiometry Answers

J Ma



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