

Name : _____ Period: _____

Properties of Water Worksheet

Word Bank:

covalent solvent deposition dissolve cohesion surface cooler universal
tension polar viscosity condensation negatively adhesion positively sublimation

1. The hydrogen and oxygen atoms are held together by covalent bonds.
2. The electrons are not shared equally creating a polar molecule.
3. The polarity of water allows it to dissolve most substances. Because of this it is referred to as the universal solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positively charged hydrogen end of one water molecule attracts the negatively charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
7. surface tension creates the skin-like surface formed due to the polar nature of water.
8. viscosity is the tendency for fluids to resist flow.
cooler water is more viscous than warmer water, allowing floating organisms to use less energy to keep from sinking.
9. condensation is when water changes from a gas to a liquid.
10. sublimation is when water changes from a solid directly to a gas.
11. deposition is when water changes from a gas directly to a solid.
12. Why does ice float?

water is less dense in solid form because hydrogen bonding between water molecules pushed molecules further apart into a set crystalline pattern

13. List the properties that ocean water has due to the fact that it contains salt.

higher density
conductivity
decreased surface tension

Reinforcement Worksheet Properties Of Water Answers

Lingjun Ying



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Table of Contents Reinforcement Worksheet Properties Of Water Answers

1. Understanding the eBook Reinforcement Worksheet Properties Of Water Answers
 - The Rise of Digital Reading Reinforcement Worksheet Properties Of Water Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Reinforcement Worksheet Properties Of Water Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Reinforcement Worksheet Properties Of Water Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reinforcement Worksheet Properties Of Water Answers
 - Personalized Recommendations
 - Reinforcement Worksheet Properties Of Water Answers User Reviews and Ratings

- Reinforcement Worksheet Properties Of Water Answers and Bestseller Lists
- 5. Accessing Reinforcement Worksheet Properties Of Water Answers Free and Paid eBooks
 - Reinforcement Worksheet Properties Of Water Answers Public Domain eBooks
 - Reinforcement Worksheet Properties Of Water Answers eBook Subscription Services
 - Reinforcement Worksheet Properties Of Water Answers Budget-Friendly Options
- 6. Navigating Reinforcement Worksheet Properties Of Water Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Reinforcement Worksheet Properties Of Water Answers Compatibility with Devices
 - Reinforcement Worksheet Properties Of Water Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reinforcement Worksheet Properties Of Water Answers
 - Highlighting and Note-Taking Reinforcement Worksheet Properties Of Water Answers
 - Interactive Elements Reinforcement Worksheet Properties Of Water Answers
- 8. Staying Engaged with Reinforcement Worksheet Properties Of Water Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reinforcement Worksheet Properties Of Water Answers
- 9. Balancing eBooks and Physical Books Reinforcement Worksheet Properties Of Water Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reinforcement Worksheet Properties Of Water Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reinforcement Worksheet Properties Of Water Answers
 - Setting Reading Goals Reinforcement Worksheet Properties Of Water Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reinforcement Worksheet Properties Of Water Answers
 - Fact-Checking eBook Content of Reinforcement Worksheet Properties Of Water Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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