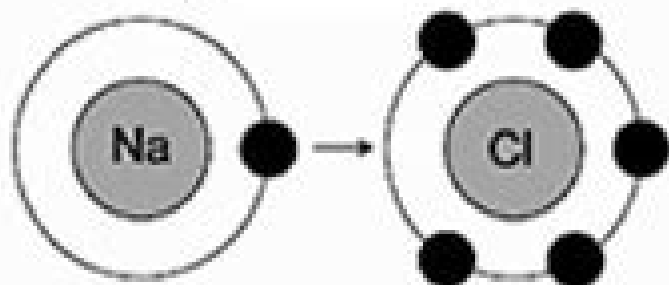
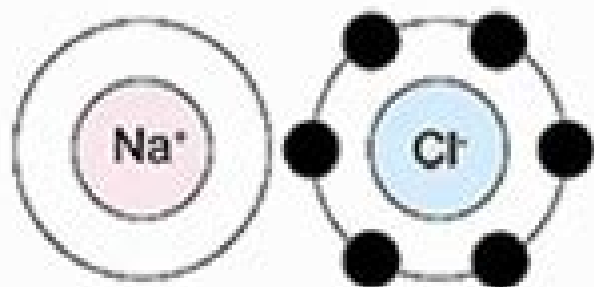


# Covalent vs. Ionic Bonds

## Ionic bond

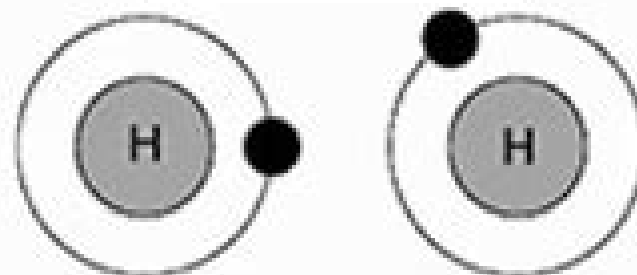


(a) The electron moves from Na to Cl

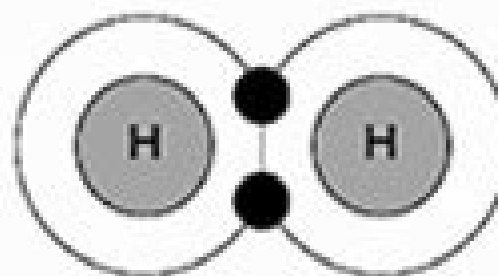


(b) Positively charged Na and negatively charged Cl attract each other and are electronically bonded together

## Covalent bond



(c) Electrons cannot move independently



(d) The atoms form bonds by sharing electrons.

# Reinforcement Ionic And Covalent Bonds Answers

**Tom Hsu PH.D.**



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