

- Which subatomic particles are located in the nucleus of a neon atom?
  - electrons and positrons
  - electrons and neutrons
  - protons and neutrons
  - protons and electrons
- Which statement best describes the nucleus of an aluminum atom?
  - It has a charge of +13 and is surrounded by a total of 10 electrons.
  - It has a charge of +13 and is surrounded by a total of 13 electrons.
  - It has a charge of -13 and is surrounded by a total of 10 electrons.
  - It has a charge of -13 and is surrounded by a total of 13 electrons.
- Which statement concerning elements is true?
  - Different elements must have different numbers of isotopes.
  - Different elements must have different numbers of neutrons.
  - All atoms of a given element must have the same mass number.
  - All atoms of a given element must have the same atomic number.
- What are the characteristics of a neutron?
  - It has no charge and no mass.
  - It has no charge and a mass of 1 amu.
  - It has a charge of +1 and no mass.
  - It has a charge of +1 and a mass of 1 amu.
- Which sequence represents a correct order of historical developments leading to the modern model of the atom?
  - the atom is a hard sphere → most of the atom is empty space → electrons exist in orbitals outside the nucleus
  - the atom is a hard sphere → electrons exist in orbitals outside the nucleus → most of the atom is empty space
  - most of the atom is empty space → electrons exist in orbitals outside the nucleus → the atom is a hard sphere
  - most of the atom is empty space → the atom is a hard sphere → electrons exist in orbitals outside the nucleus
- Which conclusion is based on the "gold foil experiment" and the resulting model of the atom?
  - An atom is mainly empty space, and the nucleus has a positive charge.
  - An atom is mainly empty space, and the nucleus has a negative charge.
  - An atom has hardly any empty space, and the nucleus has a positive charge.
  - An atom has hardly any empty space, and the nucleus has a negative charge.
- Subatomic particles can usually pass undeflected through an atom because the volume of an atom is composed of
  - an uncharged nucleus
  - largely empty space
  - neutrons
  - protons
- What is the net charge on an ion that has 9 protons, 11 neutrons, and 10 electrons?
  - 1+
  - 2+
  - 1-
  - 2-
- What is the total charge of the nucleus of a carbon atom?
  - 6
  - 0
  - +6
  - +12
- Which notation represents an atom of sodium with an atomic number of 11 and a mass number of 24?
  - $^{24}_{11}\text{Na}$
  - $^{11}_{24}\text{Na}$
  - $^{11}_{11}\text{Na}$
  - $^{24}_{11}\text{Na}$
- Compared to an atom of phosphorus-31, an atom of sulfur-32 contains
  - one less neutron
  - one less proton
  - one more neutron
  - one more proton
- Which subatomic particle has a negative charge?
  - proton
  - electron
  - neutron
  - positron
- Compared to a proton, an electron has
  - a greater quantity of charge and the same sign
  - a greater quantity of charge and the opposite sign
  - the same quantity of charge and the same sign
  - the same quantity of charge and the opposite sign
- How many electrons are in an  $\text{Fe}^{2+}$  ion.
  - 24
  - 26
  - 28
  - 56
- Which two particles make up most of the mass of a hydrogen-2 atom?
  - electron and neutron
  - electron and proton
  - proton and neutron
  - proton and positron
- Which two particles each have a mass approximately equal to one atomic mass unit?
  - electron and neutron
  - electron and positron
  - proton and electron
  - proton and neutron
- What is the mass number of an atom that has six protons, six electrons, and eight neutrons?
  - 6
  - 12
  - 14
  - 20
- What is the total number of neutrons in the nucleus of a neutral atom that has 19 electrons and a mass number of 39?
  - 19
  - 20
  - 39
  - 58
- In which list are the elements arranged in order of increasing atomic mass?
  - Cl, K, Ar
  - Fe, Co, Ni
  - Te, I, Xe
  - Ne, F, Na

# Regents Review Atomic Structure Answers

**Sasha Alcott**



## **Regents Review Atomic Structure Answers:**

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web 1  $\text{N}_2 + 2\text{O}_2 \rightarrow \text{N}_2\text{O}_4$  a if 15.0g of  $\text{N}_2\text{O}_4$  was produced how many moles of  $\text{O}_2$  were required 0.326 mol  $\text{O}_2$  b if  $4.0 \times 10^3$  moles of oxygen reacted how many grams of

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web answers to stoichiometry mole to mass problems 1 hydrogen gas can be produced through the following reaction  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  g how many grams

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web may 28 2020 the number of moles and the mass in milligrams of diatomic oxygen formed by the decomposition of 1.252 g of mercury(II) oxide the number of moles and

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web the map will help with a variety of stoichiometry problems such as mass to mass mole to mole volume to volume molecules to molecules and any combination of units they

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web 50.0 g of  $\text{N}_2$  nitrogen in problem 2  $2\text{AgNO}_3 + \text{BaCl}_2 \rightarrow 2\text{AgCl} + \text{Ba(NO}_3)_2$  how many grams of AgCl silver chloride are produced from 50.0 g of  $\text{AgNO}_3$  silver nitrate 5 how

**mole mass and mass mass problems lardbucket.org** - Jun 16 2022

web how many liters of oxygen are necessary  $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$  chemistry if8766 instructional fair inc stoichiometry mole mole problems  $\text{N}_2 + 3\text{H}_2$

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web this is a comprehensive end of chapter set of practice problems on stoichiometry that covers balancing chemical equations mole ratio calculations limiting reactants and

**5.2.1 practice problems reaction stoichiometry** - Mar 26 2023

web mar 13 2023 these mass relationships made through moles are called stoichiometry gk stoicheon element metry measure using mole and mass relationships we can

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web  $\text{g Sn} \rightarrow \text{mol Sn} \rightarrow \text{mol Hf}$  step 2 solve 75.0 g Sn  $\rightarrow 1 \text{ mol Sn} \rightarrow 118.69 \text{ g Sn} \rightarrow 2 \text{ mol Hf} \rightarrow 1 \text{ mol Sn} \rightarrow 126 \text{ mol Hf}$  step 3 think about your result the mass of tin is less than one

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web stoichiometry stoichiometry ii mole mass problems quiz to solve mole mass problems requires a balanced chemical equation and a mole ratio use the coefficients

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web the mass of the given substance is converted into moles by use of the molar mass of that substance from the periodic table then the moles of the given substance are converted

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web mole mole and mole mass problems mixed problems mole mole and mole mass worksheet challenge problem stoichiometry this semester begins with the

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web how many moles of salt are in 13.8 g  $\text{NaCl}$  text  $\text{NaCl}$  13.8 g 13.8 point 8 start text  $\text{NaCl}$  end text of sodium chloride express the answer using 3 significant figures your answer should be

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web collectively these conversions are called mole mass calculations a stoichiometry calculation converting between masses and moles of different substances in a chemical

12.3 mass mole stoichiometry chemistry libretexts - Aug 31 2023

web sep 21 2022 while the mole ratio is ever present in all stoichiometry calculations amounts of substances in the laboratory are most often measured by mass therefore

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web the beck anxiety inventory bai beck et al 1988 is a self report inventory for measuring the severity of anxiety in psychiatric populations an initial item pool of 86

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web beck anxiety inventory a 21 question self report instrument used to quantify the degree of individual anxiety and regarded as particularly useful for identifying panic each

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web with the beck anxiety inventory patients respond to 21 items rated on a scale from 0 to 3 each item is descriptive of subjective somatic or panic related symptoms of anxiety

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web the beck anxiety inventory bai and the beck depression inventory bdi are commonly used self report questionnaires to determine the presence of anxiety or

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web feb 22 2021 the structure and psychometric properties of the beck anxiety inventory bai renowned for its ability to distinguish between depression and anxiety symptoms

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web jul 4 2011 this study focuses on the beck anxiety inventory bai as a severity indicator for anxiety in primary care patients with different anxiety disorders social phobia panic

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web the beck anxiety inventory bai beck epstein brown steer 1988 beck steer 1993 is a 21 item scale developed to assess the severity of anxiety symptoms

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web an inventory for measuring clinical anxiety psychometric properties journal of consulting and clinical psychology 56 893 897 beck anxiety inventory bai below is a list of

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web may 30 2017 the beck anxiety inventory bai beck et al 1988 beck and steer 1993 is a 21 item inventory which identifies anxiety symptoms and quantifies their intensity

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web jan 1 2020 the beck anxiety inventory bai is a measure used to assess severity of anxiety beck and steer 1993 introduction the bai is a 21 item self report instrument

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web oct 1 2022 to score the beck anxiety inventory you need to add the scores on the 21 symptoms that can range between 0 and 63 points a total score of 0 7 is interpreted

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web beck anxiety inventory the beck anxiety inventory bai beck et al 1988 is a 21 item self report measure that emphasizes physiological symptoms of anxiety with high

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web nov 7 2011 the measures reviewed below include the state trait anxiety index the beck anxiety inventory and the anxiety subscale of the hospital anxiety and depression