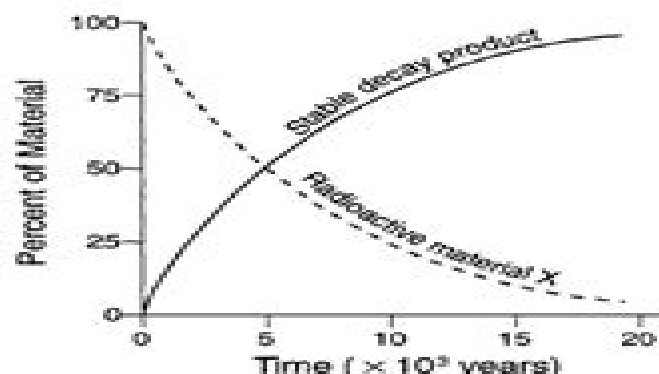


1. What fraction of a Sr-90 sample remains unchanged after 87.3 years?

- A) $\frac{1}{2}$
B) $\frac{1}{4}$
C) $\frac{1}{3}$
D) $\frac{1}{8}$

2. Base your answer to the following question on the graph below. The graph represents the decay of radioactive material X into a stable decay product.



If radioactive material X were heated, the length of its half-life period would

- A) decrease
B) increase
C) remain the same

3. Exactly how much time must elapse before 16 grams of potassium-42 decays, leaving 2 grams of the original isotope?

- A) 8×12.4 hours
B) 2×12.4 hours
C) 3×12.4 hours
D) 4×12.4 hours

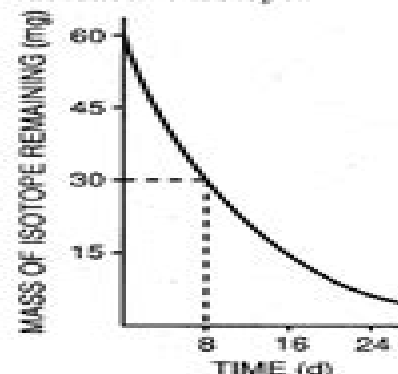
4. An original sample of K-40 has a mass of 25.00 grams. After 3.9×10^9 years, 3.125 grams of the original sample remains unchanged. What is the half-life of K-40?

- A) 1.3×10^9 y
B) 3.9×10^9 y
C) 2.6×10^9 y
D) 1.2×10^9 y

5. After 30 days, 5.0 grams of a radioactive isotope remains from an original 40.-gram sample. What is the half-life of this element?

- A) 15 days
B) 10 days
C) 20 days
D) 5 days

6. The graph below represents the decay of a radioactive isotope.



Based on Reference Table H, which radioisotope is best represented by the graph?

- A) ^{198}Au
B) ^{222}Rn
C) ^{131}I
D) ^{32}P

Regents Chemistry Half Reaction Questions

R Barnett

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Table of Contents Regents Chemistry Half Reaction Questions

1. Understanding the eBook Regents Chemistry Half Reaction Questions
 - The Rise of Digital Reading Regents Chemistry Half Reaction Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Regents Chemistry Half Reaction Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Regents Chemistry Half Reaction Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Regents Chemistry Half Reaction Questions
 - Personalized Recommendations
 - Regents Chemistry Half Reaction Questions User Reviews and Ratings
 - Regents Chemistry Half Reaction Questions and Bestseller Lists
5. Accessing Regents Chemistry Half Reaction Questions Free and Paid eBooks
 - Regents Chemistry Half Reaction Questions Public Domain eBooks
 - Regents Chemistry Half Reaction Questions eBook Subscription Services
 - Regents Chemistry Half Reaction Questions Budget-Friendly Options
6. Navigating Regents Chemistry Half Reaction Questions eBook Formats

- ePub, PDF, MOBI, and More
 - Regents Chemistry Half Reaction Questions Compatibility with Devices
 - Regents Chemistry Half Reaction Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Regents Chemistry Half Reaction Questions
 - Highlighting and Note-Taking Regents Chemistry Half Reaction Questions
 - Interactive Elements Regents Chemistry Half Reaction Questions
 8. Staying Engaged with Regents Chemistry Half Reaction Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Regents Chemistry Half Reaction Questions
 9. Balancing eBooks and Physical Books Regents Chemistry Half Reaction Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Regents Chemistry Half Reaction Questions
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Regents Chemistry Half Reaction Questions
 - Setting Reading Goals Regents Chemistry Half Reaction Questions
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Regents Chemistry Half Reaction Questions
 - Fact-Checking eBook Content of Regents Chemistry Half Reaction Questions
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
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 - Interactive and Gamified eBooks

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