

Assessment 17 Worksheet: Redox Reactions & Oxidation Numbers

Complete the following questions, and upload the file to the dropbox.

1. Why can an oxidation reaction not happen without a reduction reaction? Your response should clearly communicate your understanding of the nature of redox reactions.

(1 point)

An oxidation reaction can not happen without a reduction reaction because for an oxidation reaction there must also be a reduction happening at the same time. This is because oxidation happens because of the "compound" that causes the oxidation, this "compound" must be reduced otherwise oxidation will not occur.

2. Determine the oxidation number of the specified element in each of the following.

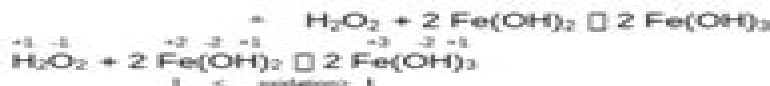
(6 points)

- a. Nitrogen in NF_3 +3
- b. Sulfur in S_8 0
- c. Chromium in CrO_4^{2-} +5
- d. Phosphorus in P_2O_5 +5
- e. Carbon in $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ 0
- f. Carbon in CHCl_3 +2

3. Determine the oxidation number for each element in the following substances. (6 points)

- a. H_2SO_3 hydrogen: +1 sulfur: +4 oxygen: -2
- b. OH^- oxygen: -2 hydrogen: +1
- c. HPO_4^{2-} hydrogen: +1 phosphorus: +5 oxygen: -2

4. Determine whether each reaction is a redox reaction by finding the oxidation numbers. Show the oxidation numbers to justify your answer. (6 points)



Redox or Not? Redox



Redox or Not? Not a Redox

Redox Reactions Chapter Assessment Answers

Priti Singhal



Redox Reactions Chapter Assessment Answers:

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