

**Syllabus reference:**

*“Write ionic half-equations and use them to construct full ionic equations.”*

**Aim of this worksheet:**

To introduce you to the idea of balancing redox equations by adjusting coefficients and by adding  $\text{OH}^-$ ,  $\text{H}_2\text{O}$ , and  $\text{H}^+$  where necessary

**Task: Balance the following redox reactions:**

- $\text{SO}_2 (\text{g}) + \text{HNO}_2 (\text{aq}) \rightarrow \text{H}_2\text{SO}_4 (\text{aq}) + \text{NO} (\text{g})$
- $\text{Al} (\text{s}) + \text{H}_2\text{SO}_4 (\text{aq}) \rightarrow \text{Al}_2(\text{SO}_4)_3 (\text{aq}) + \text{H}_2 (\text{g})$
- $\text{Au}^{3+} (\text{aq}) + \text{I}^- (\text{aq}) \rightarrow \text{Au} (\text{s}) + \text{I}_2 (\text{s})$
- $\text{S}^{2-} (\text{aq}) + \text{I}_2 (\text{s}) \rightarrow \text{SO}_4^{2-} (\text{aq}) + \text{I}^- (\text{aq})$
- $\text{Br}_2 (\text{aq}) + \text{OH}^- (\text{aq}) \rightarrow \text{Br}^- (\text{aq}) + \text{BrO}_3^- (\text{aq})$
- $\text{H}_2\text{O}_2 (\text{aq}) + \text{ClO}_4^- (\text{aq}) \rightarrow \text{O}_2 (\text{g}) + \text{ClO}_2^- (\text{aq})$
- $\text{Mn} (\text{s}) + \text{HNO}_3 (\text{aq}) \rightarrow \text{Mn}^{2+} (\text{aq}) + \text{NO}_2 (\text{g})$
- $\text{I}_2 (\text{s}) + \text{OCl}^- (\text{aq}) \rightarrow \text{IO}_3^- (\text{aq}) + \text{Cl}^- (\text{aq})$
- $\text{Cr}_2\text{O}_7^{2-} (\text{aq}) + \text{HNO}_2 (\text{aq}) \rightarrow \text{Cr}^{3+} (\text{aq}) + \text{NO}_3^- (\text{aq})$
- $\text{CrO}_4^{2-} (\text{aq}) + \text{S}^{2-} (\text{aq}) \rightarrow \text{Cr}(\text{OH})_3 (\text{s}) + \text{S} (\text{s})$

# Redox Equations Worksheet With Answers

**Lise-Lotte Österlund, Margareta  
Ekborg, Anders Berg**



## **Redox Equations Worksheet With Answers:**

*Revise A2 Chemistry for AQA* Paddy Gannon, 2005-04 Part of the series of AS and A2 revision guides this title gives students what they need to know for the AQA exams It includes material organised into bite sized chunks of information

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**Principles of Redox Reactions** Malika Ammam, 2018-03-05 As a teacher of physical chemistry I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry In this Section 3 I attempted to discuss some fundamental principles related to redox processes by focusing on the species that might lose or gain electrons determination of the oxidation numbers or states of atoms in compounds and ways of balancing redox reactions To further clarify the discussed concepts numerous questions and problems with detailed answers are provided Most of these questions are formulated by students like you I believe that this Section 3 would greatly help students with levels varying from high school to advanced university classes **Balancing Redox Equations** Lifeliqe, 2019 This lesson plan covers how to balance redox equations by inspection through two systematic processes

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**Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance)** Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these chemical reactions

**Redox Models in Chemistry** Lise-Lotte Österlund, Margareta Ekborg, Anders Berg, 2010

**Redox Reactions Explained (General Chemistry Quick Review)** E Staff, Learn and review on the go Use Quick Review Science Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students [Learning Channel](#), 1980

**Redox Reactions for Group 5 Elements, Including Element 105, in Aqueous Solutions** G. V. Ionova, Gesellschaft für Schwerionenforschung (Darmstadt), 1992

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*A Better Way to Learn Redox Chemistry Part 1: Summarized Principles and Over 250 Answered Questions and Problems* Malika Ammam, 2017-11-25 This short book part 1 summarizes some basics of redox chemistry electrochemistry in solution Each section of the book contains concise and clear referenced summaries of redox concepts Over 250 answered questions and problems are provided to further clarify the discussed principles

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**Basic Principles of Inorganic Redox Reactions in Aqueous Solutions** A.S. Ghosh Mazumdar, 1971

*Redox* J. Schüring, H.D. Schulz, W.R. Fischer, Jürgen Böttcher, W.H.M. Duijnvisveld, 2000-01-24 Few processes are as important for environmental geochemistry as the interplay

between the oxidation and reduction of dissolved and solid species The knowledge of the redox conditions is most important to predict the geochemical behaviour of a great number of components the mobilities of which are directly or indirectly controlled by redox processes The understanding of the chemical mechanisms responsible for the establishment of measurable potentials is the major key for the evaluation and sensitive interpretation of data This book is suitable for advanced undergraduates as well as for all scientists dealing with the measurement and interpretation of redox conditions in the natural environment

**Thermal Redox Reactions Between Metal Ions and Radicals in Aqueous Solutions** Zuhair Kamal Jamil,1965  
**Ab-initio Molecular Dynamics Simulation of Redox Reactions in Aqueous Solution** Jochen Blumberger,2004  
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