

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_6H_{12}O_6$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

Empirical $C_6H_{12}O_6$ (molecular)

Formula represents the proportions of _____ of
positive and negative ions in a formula
unit.

Atomic Mass

The proportions that atoms combine through
ions together.

Mass

m mass charge charge charge

m charge charge charge

Subscript

m charge charge

m charge charge charge

m charge

Note Taking Guide Episode 601 Chemical Formula

Ensheng Dong



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