

Periodic and Groups: Valence Electrons: Mastery

Periodic and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

		Groups 13-18 are sometimes referred to as the main group elements. Groups 1-10 are sometimes called transition metals. Groups 11-12 are sometimes called inner transition metals. Groups 13-18 are sometimes called the main group elements.																		
		Group 1		Groups 13-18										Group 18						
Period		1A	2A											13A	14A	15A	16A	17A	18A	
1		H																		He
2		Li	Be											B	C	N	O	F	Ne	
3		Na	Mg	Transition metals (not family members)										Al	Si	P	S	Cl	Ar	
4		K	Ca																	

Thinking like the game "Monopoly", answer that an element is in period 3 and group 1.

What element is in Group 13A and Period 2?

What group and period is Helium in?

What element is in Group 2, and Period 10A? (Group 2 element is always that in Column 2) (The element is oxygen)

What element is in Group 2A, and Period 2?

What element Helium is?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each group contains has the same number of valence electrons. Valence electrons are the electrons that are involved in chemical bonding.

Hydrogen 1



1 valence electron

Helium 2



2 valence electrons

Group 17



7 valence electrons

Group 18, including Helium and Argon, has 8 valence electrons. How many valence electrons will they have?

How many electrons?

Hydrogen and Helium are both in Group 1A (Hydrogen) 2 (Helium).

Group 18A: - Shows you more clearly that there is a full shell of electrons. Elements are in Group 18 and Helium are included in it (except 1-10). Helium is not in Group 18 (valence electrons). "He 1A, 2A" (Group 18A has a full shell of valence electrons) (18A has valence electrons 18A has 18 valence electrons).

Periods And Groups Valence Electrons Masses Answers

Lei Huang



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The Practice of Chemistry Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

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