

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.*

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.*

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.*

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- O C **Be** **Ne** Same Period
 - Na Rb **Fr** **H** Same Group
 - Pb** **C** Sn Si Same Group
 - Au W S **Fr** **Ne** Zn Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- O C **Be** **Ne** Same Period
- Na Rb **Fr** **H** Same Group
- Pb** **C** Sn Si Same Group
- Au W S **Fr** **Ne** Zn Challenge

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Annelies Wilder-Smith



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Homenatge professor Josep M. Costa (eBook) 1a part. Trends in electrochemistry and corrosion at the beginning of the 21st century Pere-Lluís Cabot Julia, Enric Brillas Coso, 2004-04-13 Esta primera parte del libro Trends in Electrochemistry and Corrosion at the beginning of the 21st century dedicado al Prof Josep M Costa en ocasi n de su 70 aniversario re ne un total de 33 art culos y revisiones originales tanto cient ficas como tecnol gicas correspondientes al campo de la Electroqu mica Estos trabajos est n escritos en espa ol e ingl s por unos 110 investigadores de todo el mundo y son exponentes representativos del gran nivel alcanzado por la investigaci n internacional en diversas materias de gran inter s en la Electroqu mica de principios de este siglo XXI Los trabajos se han agrupado en 5 cap tulos generales relativos a los campos de Electroqu mica Molecular Electrodeposici n Electrodos Modificados Descontaminaci n Electroqu mica y Sensores y Electroan lisis This first part of the book Trends in Electrochemistry and Corrosion at the beginning of the 21st century dedicated to Professor Josep M Costa in occasion of his 70th birthday collects 33 original papers and reviews both scientific and technologic corresponding to the field of Electrochemistry These works are written in English and Spanish by about 110 researchers of all around the world and are representative exponents of the high level reached by the international research in several topics of great interest in Electrochemistry at the beginning of the 21st Century The works have been distributed into five general chapters related to the fields of Molecular Electrochemistry Electrodeposition Modified Electrodes Electrochemical Depollution and Sensors and Electroanalysis *Inorganic Chemistry* Tina Overton, Jonathan Rourke, Fraser A. Armstrong, 2018 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject *Inorganic Chemistry Seventh Edition* is the ideal course companion for the duration of a student s degree The authors have drawn upon their extensive teaching and research experience to update this text the seventh edition retains the much praised clarity of style and layout from previous editions while offering an enhanced section on expanding our horizons The latest innovative applications of green chemistry have been added to clearly illustrate the real world significance of the subject This edition also sees a greater used of learning features including substantial updates to the problem solving questions additional self tests and walk through explanations which enable students to check their understanding of key concepts and develop problem solving skills Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resources *Inorganic Chemistry Seventh Edition* is accompanied by a range of online resources For registered adopters of the text DT Figures marginal structures and tables of data ready to download DT Test bank For students DT Answers to self tests and exercises from the book DT Tables for group theory DT Web links DT Links to interactive structures and other resources on www.chemtube3d.com Scientific and Technical Aerospace Reports , 1984 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA

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