## Section-A

- (a) Draw and explain the phase diagram of Zn-Mg system.
  - (b) Write an explanatory note on Enzymatic catalysis.
- 3. (a) Draw and explain the phase diagram of  $H_2$ O-system.
  - (b) Explain the concepts of promoters, inhibitors and poisoners.

## Section-B

- 4. (a) 100 ml of water sample requires 20 ml N/50 H<sub>2</sub>SO<sub>4</sub> during titration by using phenolphthalein indicator and 26 ml of same acid by using methyl orange indicator. Calculate the alkalinity of each type in terms of CaCO<sub>3</sub> equivalent.
  - (b) What do you understand by demineralization of water? Discuss in detail the ion-exchange process for demineralization of hard water with help of neat, clean and labeled diagram.
- 5. (a) A zeolite softener was 75% exhausted by removing the hardness completely when the 100000 litres of hard water sample passed through it. The exhausted zeolite bed requires 145 litres of 25% NaCl solution for its complete regeneration. Calculate the hardness of water.

# **Sample Question Paper Second Semester G Scheme**

**R Sandford** 

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