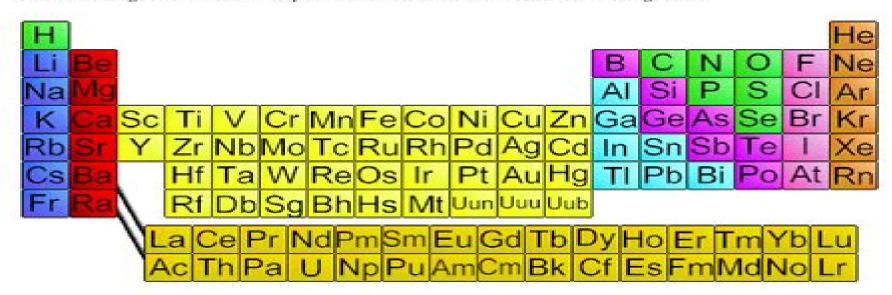


Part 1:

Follow the instructions in the Periodic Table Virtual Lab Lesson to color the periodic table.

Click on the image below -> edit -> drop down arrow beside the line -> select tool -> change colors.



Part 2: Go to chemicalelements com.

- 1. Click on Alkali Metals (left bar) to answer the following questions:
 - a. What is the group number? [1]
 - b. Are these metals reactive? [Yes, they are highly reactive.]
 - Do these metals occur freely in nature? [No, they don't occur freely in nature, but they are found in various minerals.]
 - d. How many electrons are in their outer shell? [1]
 - What are the three characteristics of ALL metals? [Shiny luster, good conductors of heat and electricity, malleable and ductile.
 - f. Are these metals soft or hard? [soft]
 - g. Name the two most reactive elements in this group? [Potassium and Sodium]
 - h. What happens when they are exposed to water? [They react vigorously with water, producing hydrogen gas and an alkaline solution. The reaction can be violent and can even ignite the hydrogen gas produced...]

Periodic Table Grouping Lab Answer Key

Yi-Tong Ma

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Table of Contents Periodic Table Grouping Lab Answer Key

- 1. Understanding the eBook Periodic Table Grouping Lab Answer Key
 - The Rise of Digital Reading Periodic Table Grouping Lab Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Grouping Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Grouping Lab Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Grouping Lab Answer Key
 - Personalized Recommendations
 - Periodic Table Grouping Lab Answer Key User Reviews and Ratings
 - Periodic Table Grouping Lab Answer Key and Bestseller Lists

- 5. Accessing Periodic Table Grouping Lab Answer Key Free and Paid eBooks
 - Periodic Table Grouping Lab Answer Key Public Domain eBooks
 - Periodic Table Grouping Lab Answer Key eBook Subscription Services
 - Periodic Table Grouping Lab Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Grouping Lab Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Table Grouping Lab Answer Key Compatibility with Devices
 - Periodic Table Grouping Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Grouping Lab Answer Key
 - Highlighting and Note-Taking Periodic Table Grouping Lab Answer Key
 - Interactive Elements Periodic Table Grouping Lab Answer Key
- 8. Staying Engaged with Periodic Table Grouping Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Grouping Lab Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Grouping Lab Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Table Grouping Lab Answer Key
- 10. Overcoming Reading Challenges
 - $\circ\,$ Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Grouping Lab Answer Key
 - Setting Reading Goals Periodic Table Grouping Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Grouping Lab Answer Key
 - Fact-Checking eBook Content of Periodic Table Grouping Lab Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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