

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

Reaction Spontaneity Study Guide Key

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Table of Contents Reaction Spontanetity Study Guide Key

1. Understanding the eBook Reaction Spontanetity Study Guide Key
 - The Rise of Digital Reading Reaction Spontanetity Study Guide Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Reaction Spontanetity Study Guide Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Spontanetity Study Guide Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reaction Spontanetity Study Guide Key
 - Personalized Recommendations
 - Reaction Spontanetity Study Guide Key User Reviews and Ratings
 - Reaction Spontanetity Study Guide Key and Bestseller Lists

5. Accessing Reaction Spontanetity Study Guide Key Free and Paid eBooks
 - Reaction Spontanetity Study Guide Key Public Domain eBooks
 - Reaction Spontanetity Study Guide Key eBook Subscription Services
 - Reaction Spontanetity Study Guide Key Budget-Friendly Options
6. Navigating Reaction Spontanetity Study Guide Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Reaction Spontanetity Study Guide Key Compatibility with Devices
 - Reaction Spontanetity Study Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Reaction Spontanetity Study Guide Key
 - Interactive Elements Reaction Spontanetity Study Guide Key
8. Staying Engaged with Reaction Spontanetity Study Guide Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Spontanetity Study Guide Key
9. Balancing eBooks and Physical Books Reaction Spontanetity Study Guide Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Spontanetity Study Guide Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Reaction Spontanetity Study Guide Key
 - Setting Reading Goals Reaction Spontanetity Study Guide Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reaction Spontanetity Study Guide Key
 - Fact-Checking eBook Content of Reaction Spontanetity Study Guide Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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