# Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	o	linear	180	180°
3	0	trigonal planar	O Jaser	120°
	1	bent	-8-0	<120°
4	o	tetrahedral	355	109.5°
	1	trigonal pyramid	3000	<109.5° (~107°)
	2	bent	- S 100.5"	<109.5°(~105°)
5	o	trigonal bipyramidal	13-38-5	90°, 120°
	1	see-saw		<90°, <120°
	2	T-structure	3	<90°
	3	linear	( ) I ( ) ( ) ( ) ( ) ( )	180°
6	o	octahehral	833.00	90°, 90°
	1	square pyramidal	33.90	90°, <90°
	2	square planar	8:3:9:9:	90°

# **Shapes Of Molecules Worksheet If8766**

**CH Cherryholmes** 

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