

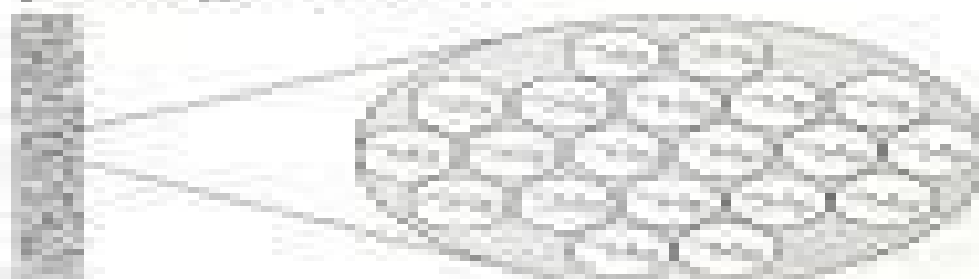
Average Atomic Mass

How can the masses of the particles within an atom be determined?

Objectives

After observing three gases, three ions, and naturally occurring three isotopes, the masses of these particles, ions, and isotopes should be determined. Comparison of particles including those belonging to the same element should show three different values resulting from isotopes of particles and elements having three different particles, neutrons. For which mass is your value the greatest value? For which element is the lightest element having three different particles? How would you determine the mass of a particle?

Model 1 - A Strip of Magnesium Metal



1. What are the masses of the particles within the metal Mg shown in Model 1?
They are the same, they are all 24.
2. What are the masses of the particles within the metal Mg shown in Model 1?
24, 24, 24.
3. How are the masses of the particles in Model 1? Are the masses of the particles the same?
Yes, they are the same, they are all 24.
4. How are the masses of the particles in Model 1? Are the masses of the particles the same?
Yes, they are the same, they are all 24.

	^{24}Mg	^{25}Mg	^{26}Mg
Number of atoms	10	5	5

5. What are the masses of the particles in the metal Mg shown in Model 1?
24, 24, 24.
6. What are the masses of the particles in the metal Mg shown in Model 1? Are the masses of the particles the same?
Yes, they are the same, they are all 24.

Periodicity Of Elements Pogil Key

M. Satake



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