

What are Acids, Bases, and Salts?

The Swedish chemist Svante Arrhenius introduced the theory of ionization and used this theory to explain much about the behavior of acids and bases.

- An Arrhenius acid is defined as any compound that dissociates in aqueous solution to form H^+ ions.



- An Arrhenius base is defined as any compound that dissociates in aqueous solution to form OH^- ions.



- Salts are compounds that dissociate in aqueous solution releasing neither H^+ ions nor OH^- ions.



Using the **Arrhenius definition**, classify the following examples as acids, bases, or salts:

HBr	<u>acid</u>	KCl	<u>salt</u>
Mg(OH) ₂	<u>base</u>	H ₃ PO ₄	<u>acid</u>
HCl	<u>acid</u>	HClO	<u>acid</u>
KNO ₃	<u>salt</u>	Al(OH) ₃	<u>base</u>
HFO ₄	<u>acid</u>	KC ₂ H ₃ O ₂	<u>salt</u>
Ba(OH) ₂	<u>base</u>	NaCl	<u>salt</u>

Note Taking Worksheet Acids Bases And Salts

Daniela Niemeyer



Note Taking Worksheet Acids Bases And Salts:

Enjoying the Melody of Phrase: An Psychological Symphony within **Note Taking Worksheet Acids Bases And Salts**

In some sort of taken by screens and the ceaseless chatter of instant interaction, the melodic splendor and emotional symphony developed by the written word frequently fade in to the back ground, eclipsed by the persistent sound and disruptions that permeate our lives. But, located within the pages of **Note Taking Worksheet Acids Bases And Salts** a wonderful literary prize overflowing with organic feelings, lies an immersive symphony waiting to be embraced. Crafted by an elegant composer of language, that interesting masterpiece conducts viewers on an emotional journey, well unraveling the concealed tunes and profound impact resonating within each carefully crafted phrase. Within the depths of the emotional analysis, we can investigate the book is main harmonies, analyze its enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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