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reaction in acid solution using the half reaction method $8 \text{H}_2\text{O}_2 + \text{Cr}_2\text{O}_7^{2-} + 2 \text{H}^+ \rightarrow 2 \text{Cr}^{3+} + 10 \text{H}_2\text{O} + 11 \text{PbO}_2 + \text{IO}_3^- + 2 \text{H}^+ + 12 \text{H}^+$
 as $\text{NO}_3^- + \text{IO}_3^- + \text{H}_2\text{AsO}_4^-$

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 + 2\text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{HNO}_2$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and H_2O b HNO_2 and $\text{C}_6\text{H}_8\text{O}_6$ b H_2O and $\text{C}_6\text{H}_8\text{O}_6$ h and HNO_2 c h and

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