

Answer all questions on the paper/answer book provided.  
A scientific calculator may be used.

**QUESTION 1**

There are four possible options for each answer in the following questions. Each question has only ONE correct answer. Choose the correct answer and write only A, B, C or D next to the question number.

- 1.1 A compound consists of ions  $X^{2+}$  and  $Y^{3-}$ . A possible formula for the compound could be:
- A  $3X2Y$
  - B  $X_2Y_3$
  - C  $X_3Y_2$
  - D  $(XY)_6$
- (2)
- 1.2 Which of the following is not a characteristic of a physical change:
- A no new products are formed
  - B changes can be reversed
  - C observable with the naked eye
  - D the parts of the substance lose their chemical identity
- (2)
- 1.3 Ionic bonding occurs between:
- A two non-metals
  - B two metals
  - C a metal and a non-metal
  - D a metal and a semi-metal
- (2)
- 1.4 Khumalo heats chocolate in a double boiler and observes the changes. The most accurate conclusion to his experiment is:
- A Chocolate undergoes a chemical change when it is heated
  - B Chocolate undergoes a physical change when it is heated
  - C Chocolate decomposes into its constituent components when heated
  - D Chocolate gives off latent energy when it melts
- (2)

# Physical Science Control Test Term 3

**Mike Jess**



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