

Name: _____ () Class: _____ Date: _____

The Periodic TableGeneral Properties of the Periodic Table

1. The Periodic Table is a table that groups elements according to certain physical properties or chemical properties.
2. The elements are arranged in order of increasing proton number (atomic number).
3. A **HORIZONTAL** row of elements is known as a Period.
Across a period, the element changes from **metal** character to **non-metal** character.
Elements on left hand side of Periodic Table are _____.
Elements on right hand side of Periodic Table are _____.
4. A **VERTICAL** row of elements is known as a Group.
5. Hydrogen and Helium are in the same Period 1.
They all have _____ electron shell.

Draw out the electronic structure of Hydrogen and Helium below:



Hydrogen



Helium

6. The following elements are in the same Period 2.
They all have _____

Draw out the electronic structure of these elements.

--	--	--	--	--	--	--	--

Lithium

Beryllium

Boron

Carbon

Nitrogen

Oxygen

Fluorine

Neon

7. Lithium is in Group 1. Lithium has _____ outer electron.
Beryllium is in Group 2. Beryllium has _____ outer electrons.
Oxygen is in Group 6. Oxygen has _____

Quick Check

Without reference to the Periodic Table, predict the Period and Group of the elements below:

- | | | | |
|------------------|-------|----------------------|-------|
| a) Ca – 2.8.8.2 | _____ | d) Xe 2.8.18.18.8 | _____ |
| b) S – 2.8.6 | _____ | e) Pb 2.8.18.32.18.4 | _____ |
| c) Ga – 2.8.18.3 | _____ | f) Rb – 2.8.18.8.1 | _____ |

Periodic Table The Science Spot Answer Key

Siddappa N.Byrareddy



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Table of Contents Periodic Table The Science Spot Answer Key

1. Understanding the eBook Periodic Table The Science Spot Answer Key
 - The Rise of Digital Reading Periodic Table The Science Spot Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table The Science Spot Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table The Science Spot Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table The Science Spot Answer Key
 - Personalized Recommendations
 - Periodic Table The Science Spot Answer Key User Reviews and Ratings

- Periodic Table The Science Spot Answer Key and Bestseller Lists
- 5. Accessing Periodic Table The Science Spot Answer Key Free and Paid eBooks
 - Periodic Table The Science Spot Answer Key Public Domain eBooks
 - Periodic Table The Science Spot Answer Key eBook Subscription Services
 - Periodic Table The Science Spot Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table The Science Spot Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table The Science Spot Answer Key Compatibility with Devices
 - Periodic Table The Science Spot Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table The Science Spot Answer Key
 - Highlighting and Note-Taking Periodic Table The Science Spot Answer Key
 - Interactive Elements Periodic Table The Science Spot Answer Key
- 8. Staying Engaged with Periodic Table The Science Spot Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table The Science Spot Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table The Science Spot Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table The Science Spot Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table The Science Spot Answer Key
 - Setting Reading Goals Periodic Table The Science Spot Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table The Science Spot Answer Key
 - Fact-Checking eBook Content of Periodic Table The Science Spot Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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