

Percent Yield Practice

1. Calculate the theoretical yield of calcium oxide if 24.8 g of calcium carbonate decomposes. Calculate the percent yield if you got 13.1 g in the laboratory.



$$\frac{24.8 \text{ g CaCO}_3}{1} \cdot \frac{1 \text{ mol CaCO}_3}{100.09 \text{ g CaCO}_3} \cdot \frac{1 \text{ mol CaO}}{1 \text{ mol CaCO}_3} \cdot \frac{56.08 \text{ g CaO}}{1 \text{ mol CaO}} = 13.9 \text{ g CaO} \quad \text{TY}$$

$$\text{PY} = \frac{\text{AY}}{\text{TY}} \cdot 100 = \frac{13.1}{13.9} \cdot 100 = \boxed{94.24\%} \Rightarrow \boxed{94.2\%}$$

2. What is the percent yield if you experimentally produce 3.65 g of copper when 1.87 g of aluminum reacts with 9.65 g copper (II) sulfate?



$$\frac{1.87 \text{ g Al}}{1} \cdot \frac{1 \text{ mol Al}}{26.98 \text{ g Al}} \cdot \frac{3 \text{ mol Cu}}{2 \text{ mol Al}} \cdot \frac{159.62 \text{ g CuSO}_4}{1 \text{ mol CuSO}_4} = 16.6 \text{ g CuSO}_4 \quad \text{Used}$$

$$\frac{9.65 \text{ g CuSO}_4}{159.62 \text{ g CuSO}_4} \cdot \frac{1 \text{ mol CuSO}_4}{1 \text{ mol CuSO}_4} \cdot \frac{3 \text{ mol Cu}}{2 \text{ mol CuSO}_4} \cdot \frac{63.55 \text{ g Cu}}{1 \text{ mol Cu}} = 3.84 \text{ g Cu}$$

$$\text{PY} = \frac{\text{AY}}{\text{TY}} \cdot 100 = \frac{3.65}{3.84} \cdot 100 = \boxed{95.05\%} \quad \text{95.1\%}$$

3. If you start with 5.0 g of C_3H_8 and react it with oxygen, what is the theoretical yield of water? If you have a percent yield of 75.0%, What is the actual yield of water?



$$\frac{5.0 \text{ g C}_3\text{H}_8}{44.11 \text{ g C}_3\text{H}_8} \cdot \frac{1 \text{ mol C}_3\text{H}_8}{1 \text{ mol C}_3\text{H}_8} \cdot \frac{4 \text{ mol H}_2\text{O}}{1 \text{ mol C}_3\text{H}_8} \cdot \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = 8.2 \text{ g H}_2\text{O} \quad \text{TY}$$

$$\text{PY} = \frac{\text{AY}}{\text{TY}} \cdot 100$$

$$\frac{\text{PY} \cdot \text{TY}}{100} = \text{AY} = \frac{75 \cdot 8.2}{100} = \boxed{6.2 \text{ g H}_2\text{O}} \quad \text{AY}$$

4. What is the theoretical yield of sodium oxide if you start with 20 g of calcium oxide?



$$\frac{20 \text{ g CaO}}{56.08 \text{ g CaO}} \cdot \frac{1 \text{ mol CaO}}{1 \text{ mol CaO}} \cdot \frac{1 \text{ mol Na}_2\text{O}}{1 \text{ mol CaO}} \cdot \frac{61.98 \text{ g Na}_2\text{O}}{1 \text{ mol Na}_2\text{O}} = \boxed{20 \text{ g Na}_2\text{O}}$$

Percent Yield Guided Problems And Answers

David Burghes, Jodie Hunter



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *General Chemistry, Reactions First* Kevin Revell, 2024-12-04 Revell's General Chemistry empowers students to grasp essential topics and concepts with more ease Using a friendly approach the text uses metaphors and relatable examples to demystify even the most challenging subjects in general chemistry *Guide to Food Safety and Quality during Transportation* John M. Ryan, 2017-05-19 Guide to Food Safety and Quality during Transportation Controls Standards and Practice Second Edition provides a solid foundation outlining logistics and delivery control solutions to protect the food transportation industry Since its first publication the U S FDA has finalized a number of Food Safety Modernization Act rules designed to improve the protection of the public from adulterants known to cause illness and death Food shippers carriers and receivers throughout the world are impacted as import controls have tightened This book provides the information needed to comply with the Act's requirements and tactics on how to achieve safety in the food supply chain Filled with legal liability and practical solutions food transporters and buyers will be able to structure company wide business practices as part of their overall food safety and quality agendas For food safety and quality students the book provides much needed insight into a critical but overlooked aspect of the food safety and food quality spectrums This food transporter piece of the overall food safety and quality puzzle provides the linking mechanism needed to improve the supply chain communication and interdependence sought after by governmental and industry executives Includes important information on how to comply with the Food Safety Modernization Act Includes technological advances in sanitation testing and traceability and highlights cost effective solutions to enhance food safety Provides practical solutions to transportation problems including container sanitation temperature controls traceability adulteration and other food safety and quality issues Presents potential sources of adulteration both chemical and biological at producer level both domestic and foreign to reduce transporter liability Provides new and updated information including environmental monitoring statistical control systems supply chain management and more *General, Organic, and Biological Chemistry* Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration **Introductory Chemistry** Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of

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Table of Contents Percent Yield Guided Problems And Answers

1. Understanding the eBook Percent Yield Guided Problems And Answers
 - The Rise of Digital Reading Percent Yield Guided Problems And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Percent Yield Guided Problems And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Percent Yield Guided Problems And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Percent Yield Guided Problems And Answers
 - Personalized Recommendations

- Percent Yield Guided Problems And Answers User Reviews and Ratings
- Percent Yield Guided Problems And Answers and Bestseller Lists
- 5. Accessing Percent Yield Guided Problems And Answers Free and Paid eBooks
 - Percent Yield Guided Problems And Answers Public Domain eBooks
 - Percent Yield Guided Problems And Answers eBook Subscription Services
 - Percent Yield Guided Problems And Answers Budget-Friendly Options
- 6. Navigating Percent Yield Guided Problems And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Percent Yield Guided Problems And Answers Compatibility with Devices
 - Percent Yield Guided Problems And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Percent Yield Guided Problems And Answers
 - Highlighting and Note-Taking Percent Yield Guided Problems And Answers
 - Interactive Elements Percent Yield Guided Problems And Answers
- 8. Staying Engaged with Percent Yield Guided Problems And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Percent Yield Guided Problems And Answers
- 9. Balancing eBooks and Physical Books Percent Yield Guided Problems And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Percent Yield Guided Problems And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Percent Yield Guided Problems And Answers
 - Setting Reading Goals Percent Yield Guided Problems And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Percent Yield Guided Problems And Answers
 - Fact-Checking eBook Content of Percent Yield Guided Problems And Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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