

| Element | symbol | Atomic number | Electronic arrangement in different orbit | | | | | |
|------------|--------|---------------|---|---|---|---|---|---|
| | | | K | L | M | N | O | P |
| Hydrogen | H | 1 | 1 | | | | | |
| Helium | He | 2 | 2 | | | | | |
| Lithium | Li | 3 | 2 | 1 | | | | |
| Beryllium | Be | 4 | 2 | 2 | | | | |
| Boron | B | 5 | 2 | 3 | | | | |
| Carbon | C | 6 | 2 | 4 | | | | |
| Nitrogen | N | 7 | 2 | 5 | | | | |
| Oxygen | O | 8 | 2 | 6 | | | | |
| Fluorine | F | 9 | 2 | 7 | | | | |
| Neon | Ne | 10 | 2 | 8 | | | | |
| Sodium | Na | 11 | 2 | 8 | 1 | | | |
| Magnesium | Mg | 12 | 2 | 8 | 2 | | | |
| Aluminium | Al | 13 | 2 | 8 | 3 | | | |
| Silicon | Si | 14 | 2 | 8 | 4 | | | |
| Phosphorus | P | 15 | 2 | 8 | 5 | | | |
| Sulphur | S | 16 | 2 | 8 | 6 | | | |
| Chlorine | Cl | 17 | 2 | 8 | 7 | | | |
| Argon | Ar | 18 | 2 | 8 | 8 | | | |
| Potassium | K | 19 | 2 | 8 | 8 | 1 | | |
| Calcium | Ca | 20 | 2 | 8 | 8 | 2 | | |

● $c = v\lambda$

● $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[c = velocity of light in free space,

v = frequency, λ = wavelength]

[\bar{v} = wave number]

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S Ashworth



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