

# Common Polyatomic Ions

acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
ammonium	$\text{NH}_4^+$
arsenate	$\text{AsO}_4^{3-}$
arsenite	$\text{AsO}_3^{3-}$
azide	$\text{N}_3^-$
benzoate	$\text{C}_7\text{H}_5\text{O}_2^-$
borate	$\text{BO}_3^{3-}$
bromate	$\text{BrO}_3^-$
carbonate	$\text{CO}_3^{2-}$
chlorate	$\text{ClO}_3^-$
chlorite	$\text{ClO}_2^-$
chromate	$\text{CrO}_4^{2-}$
cyanide	$\text{CN}^-$
dichromate	$\text{Cr}_2\text{O}_7^{2-}$
dihydrogen phosphate	$\text{H}_2\text{PO}_4^-$
dihydrogen phosphite	$\text{H}_2\text{PO}_3^-$
hydrogen carbonate	$\text{HCO}_3^-$
hydrogen phosphate	$\text{HPO}_4^{2-}$
hydrogen phosphite	$\text{HPO}_3^{2-}$
hydrogen sulfate	$\text{HSO}_4^-$
hydrogen sulfide	$\text{HS}^-$
hydrogen sulfite	$\text{HSO}_3^-$
hydroxide	$\text{OH}^-$
hypochlorite	$\text{ClO}^-$
iodate	$\text{IO}_3^-$
manganate	$\text{MnO}_4^{2-}$
nitrate	$\text{NO}_3^-$
nitrite	$\text{NO}_2^-$
oxalate	$\text{C}_2\text{O}_4^{2-}$
perchlorate	$\text{ClO}_4^-$
permanganate	$\text{MnO}_4^-$
peroxide	$\text{O}_2^{2-}$
phosphate	$\text{PO}_4^{3-}$
phosphite	$\text{PO}_3^{3-}$
silicate	$\text{SiO}_3^{2-}$
sulfate	$\text{SO}_4^{2-}$
sulfite	$\text{SO}_3^{2-}$
tartrate	$\text{C}_4\text{H}_4\text{O}_6^{2-}$
thiocyanate	$\text{SCN}^-$
thiosulfate	$\text{S}_2\text{O}_3^{2-}$

$\text{AsO}_3^{3-}$	arsenite
$\text{AsO}_4^{3-}$	arsenate
$\text{BO}_3^{3-}$	borate
$\text{BrO}_3^-$	bromate
$\text{C}_2\text{H}_3\text{O}_2^-$	acetate
$\text{C}_2\text{O}_4^{2-}$	oxalate
$\text{C}_4\text{H}_4\text{O}_6^{2-}$	tartrate
$\text{C}_7\text{H}_5\text{O}_2^-$	benzoate
$\text{ClO}^-$	hypochlorite
$\text{ClO}_2^-$	chlorite
$\text{ClO}_3^-$	chlorate
$\text{ClO}_4^-$	perchlorate
$\text{CN}^-$	cyanide
$\text{CO}_3^{2-}$	carbonate
$\text{Cr}_2\text{O}_7^{2-}$	dichromate
$\text{CrO}_4^{2-}$	chromate
$\text{H}_2\text{PO}_3^-$	dihydrogen phosphite
$\text{H}_2\text{PO}_4^-$	dihydrogen phosphate
$\text{HCO}_3^-$	hydrogen carbonate
$\text{HPO}_3^{2-}$	hydrogen phosphite
$\text{HPO}_4^{2-}$	hydrogen phosphate
$\text{HS}^-$	hydrogen sulfide
$\text{HSO}_3^-$	hydrogen sulfite
$\text{HSO}_4^-$	hydrogen sulfate
$\text{IO}_3^-$	iodate
$\text{MnO}_4^-$	permanganate
$\text{MnO}_4^{2-}$	manganate
$\text{N}_3^-$	azide
$\text{NH}_4^+$	ammonium
$\text{NO}_2^-$	nitrite
$\text{NO}_3^-$	nitrate
$\text{O}_2^{2-}$	peroxide
$\text{OH}^-$	hydroxide
$\text{PO}_3^{3-}$	phosphite
$\text{PO}_4^{3-}$	phosphate
$\text{S}_2\text{O}_3^{2-}$	thiosulfate
$\text{SCN}^-$	thiocyanate
$\text{SiO}_3^{2-}$	silicate
$\text{SO}_3^{2-}$	sulfite
$\text{SO}_4^{2-}$	sulfate

# Physical Science Formulas With Polyatomic Ions Answer

**Victor M. Corman**



## **Physical Science Formulas With Polyatomic Ions Answer:**

## Unveiling the Magic of Words: A Review of "**Physical Science Formulas With Polyatomic Ions Answer**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Physical Science Formulas With Polyatomic Ions Answer**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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