

Name: _____ KEY _____ Period _____ Date: _____

Fill in the blank (2 points each blank)

1. Elements that have the properties of both metal and non-metals are metalloids.
2. What do elements in the same group have the same number of? Valence electrons
3. Which element would have the highest electron affinity? Fluorine
4. What element is at period 3, group 8A? Argon
5. How many valence electrons does Bromine have? 7
6. The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
7. Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
8. Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity
9. Which element would have the smallest atomic radius? Helium
10. Which element would have the highest ionization energy? Helium
11. For the questions below and based on the information provided, fill in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Mass Number
Potassium	K	19	19	22	19	41
Phosphorous	P	15	15	18	15	33
Beryllium	Be	4	4	4	4	8
Sulfur	S	16	16	20	16	36

12. Draw Bohr Models for (3 points each)

a. Na23

b. N15

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Alien Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

Periodic Table Test A Answer Key

The Princeton Review



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