

Periodic Table of the Elements

KEY

Atomic Mass → 12.011 ← Selected Oxidation States
Symbol → C
Atomic Number → 6
Electron Configuration → 2-4

Relative atomic masses are based on $^{12}\text{C} = 12$ (exact)
Notes: Numbers in parentheses are mass numbers of the most stable or common isotope.

Period	1	2	Group										13	14	15	16	17	18
1	H	He																
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg											Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo

Lanthanide Series: Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu

Actinide Series: Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, No, Lr

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Periodic Table Reference Sheet

Richard Bailey



Periodic Table Reference Sheet:

Unveiling the Energy of Verbal Art: An Psychological Sojourn through **Periodic Table Reference Sheet**

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<https://crm.avenza.com/public/uploaded-files/index.jsp/Recipe%20To%20Make%20Pastry%20For%20Tarte.pdf>

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