

Anatomy and Physiology (Sample Questions- Chapter 2)

Name of the Student:

Question 1. Sodium has an atomic number of 11 and an atomic mass of 23. Sodium has _____.

- A. 12 neutrons and 11 protons**
- B. 12 protons and 11 neutrons
- C. 12 electrons and 11 neutrons
- D. 12 protons and 11 electrons
- E. 12 electrons and 11 protons

Question 2. What type of bond attracts one water molecule to another?

- A. An ionic bond
- B. A peptide bond
- C. A hydrogen bond**
- D. A covalent bond
- E. A hydrolytic bond

Question 3. Which of these has the highest H^+ concentration?

- A. Lemon juice, pH = 2.3**
- B. Red wine, pH = 3.2
- C. Tomato juice, pH = 4.7
- D. Saliva, pH = 6.6
- E. Household ammonia, pH = 10.8

Ques 4. An atom is made up of protons, neutrons, and electrons. The proton carries the following charge:

- A. negative
- B. no charge
- C. positive**
- D. none of the above

Question 5. In the following reaction, what is(are) the product(s)? $CO_2 + H_2O \rightarrow H_2CO_3$

- A. H_2CO_3**
- B. CO_2 and H_2O
- C. CO_2 and H_2CO_3
- D. H_2O and H_2CO_3

Sample Questions Chapter With Answers

Arshad Iqbal



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Table of Contents Sample Questions Chapter With Answers

1. Understanding the eBook Sample Questions Chapter With Answers
 - The Rise of Digital Reading Sample Questions Chapter With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Questions Chapter With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Questions Chapter With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Questions Chapter With Answers
 - Personalized Recommendations
 - Sample Questions Chapter With Answers User Reviews and Ratings

- Sample Questions Chapter With Answers and Bestseller Lists
- 5. Accessing Sample Questions Chapter With Answers Free and Paid eBooks
 - Sample Questions Chapter With Answers Public Domain eBooks
 - Sample Questions Chapter With Answers eBook Subscription Services
 - Sample Questions Chapter With Answers Budget-Friendly Options
- 6. Navigating Sample Questions Chapter With Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Questions Chapter With Answers Compatibility with Devices
 - Sample Questions Chapter With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Questions Chapter With Answers
 - Highlighting and Note-Taking Sample Questions Chapter With Answers
 - Interactive Elements Sample Questions Chapter With Answers
- 8. Staying Engaged with Sample Questions Chapter With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Questions Chapter With Answers
- 9. Balancing eBooks and Physical Books Sample Questions Chapter With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Questions Chapter With Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sample Questions Chapter With Answers
 - Setting Reading Goals Sample Questions Chapter With Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sample Questions Chapter With Answers
 - Fact-Checking eBook Content of Sample Questions Chapter With Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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