

Directions: For each of the following, write the letter of the term or phrase that best completes the sentence.

1. In a solution of sugar and water, the water is the _____.
a. solvent
b. solute
2. In the dissolving process, negative polar ends are attracted to _____.
a. negative
b. positive
3. Grinding a solute increases its surface area, making it dissolve more _____.
a. slowly
b. quickly
4. A gas's solubility is best in a liquid solvent when the solution is under _____ pressure.
a. high
b. low
5. Different substances have _____ solubilities.
a. different
b. the same
6. The concentration of a solution of two or more liquids is expressed as a percentage by _____.
a. area
b. volume
7. As the temperature of a liquid solvent increases, the amount of solid solute that can be dissolved in the solvent usually _____.
a. increases
b. decreases
8. Lines on a graph that show how much solute a solvent can hold at a given temperature are called _____.
a. solution graph
b. solubility curves
9. In the ionization process, water pulls the molecules of a polar substance apart into _____.
a. crystals
b. ions
10. If a solute is a(n) _____, the solution can conduct electricity.
a. electrolyte
b. nonelectrolyte
11. The more particles of a solute that are added to a solution, the _____ the freezing point of the solution.
a. lower
b. higher

Sections 1, 2, and 3 (page 20)

1. a
2. b
3. b
4. a
5. a
6. b
7. a
8. b
9. b
10. a
11. a

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Holland, James Gordon Holland, 1976 **Nursing Model Question Paper (Part 12) - 2024** Akash Tiwari, 2024-01-08

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