

Section 18.2 Reversible Reactions and Equilibrium

■ OBJECTIVES

- Describe how the *amounts of reactants and products change* in a chemical system at equilibrium.
- Identify three *stresses* that can change the equilibrium position of a chemical system.
- Explain what the *value of K_{eq}* indicates about the position of equilibrium.

Section 18 2 Reversible Reactions And Equilibrium Answers

AW Chickering



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