

STOICHIOMETRY

This section explains how to calculate the masses of neutrons required or produced, based on a stoichiometric process. It teaches you how to interpret chemical equations in terms of interacting nuclei, representative particles, moles, and mass values of NTP.

1. How can you determine the quantities of reactants and products in a chemical reaction?

*Source: Authors' estimates. χ^2 test for heterogeneity: $\chi^2 = 1.0$, $df = 1$, $p = 0.32$.

21. Quantity usually denotes the _____ **amount** of a substance expressed in grams or moles.
22. A hex-kinase is an enzyme found in 3 isoforms (HK1, 2 and 3), 1 type (HK1, 2 and 3), and 4 types (HK1, 2 and 3). "Purification" is "indicated separation" from the concentration of HK1, 2 and 3.

$$2000 + 2000 + 1 + 1 + 1 + 1 = 8004 \text{ lbs.}$$

4. In the following sentence, underline the subject and circle the predicate of the sentence. 10-11

86. *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.) *Chloroceryle alpestris* (Aud.)

10. **Explain the difference between a "strong" and a "weak" argument.**

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7. Consider the letters of the words that tell us what kind of information your **DATA SET** got from a chemical reaction.

- a. λ - wavelength
- b. ν - frequency
- c. n - index of refraction
- d. ν - number of particles
- e. λ - size of particles

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Saraswati Experts



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-II H.K. Dass, Rama Verma & Bhagwat S. Sharma, S Chand's Mathematics books for Classes IX and X are completely based
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Mathematical Reviews ,1981

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