

Bio 2

Section 1: Chemical bonds

There are 3 types of chemical bonds covalent, ionic, and hydrogen.

Covalent bonds are strong bonds that occur between atoms in the same nucleus

Ionic bonds are moderately strong, occurs between charged atoms

Electrons transfer from one atom to another. It creates strong electrostatic attraction between oppositely charged ions

Hydrogen bond are weak and occurs between atoms in different molecules or parts of molecules. The attractions between areas of molecules with opposite partial charge

The glue that holds covalently bonded atoms together is one or more electrons that are shared between 2 atoms

Electrons transfer in an ionic bond

If two atoms both lose one electron each, they will not form an ionic bond with each other because they will have the same charge

Electrons are shared in a covalent bond

Hydrogen bonds attract molecules or parts of molecules to each other

Electrons are smaller than protons and neutrons

Protons and neutrons are found in the nucleus

Ionic bonds occur between 2 opposite charged ions

Hydrogen bonds have weak interactions

Covalent bonds involve sharing of electrons between atoms

There are 4 properties of water

1. Water is an excellent solvent and can dissolve a wide range of substances

Section 2 Study Guide Types Of Bonds Chapter

Lei Huang



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