

Acids and Bases Worksheet

 Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases	
Conductivity in Aqueous Solution	very	very	
Taste	Sove	Bitter	
Feel		Slippery	
Reaction with Metal	produces H2 gas		
Reaction with Carbonate	produces Cb2 gas		
Colour on Litmus Paper	Red	Blue.	
Colour with Universal Indicator	Red	Purple.	
Colour with phenolphthalein	Colouriess	Dink.	

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO ₃	A	nitric acid
Mg(OH)₂	_B_	magnesium hydroxide
H ₃ PO ₄	_A	Phosphoric acid
H ₂ CO ₃	A	. Chroonic acid
HF	A	hydrofluoric acid
Fe(OH) ₃	B	iron (111) hydroxide.

3. Write the formula for each of the following acids/bases:

hydrochloric acid	HCI_	hydrochloric acid	HCI
aluminum hydroxide	AI(OH)3	sodium hydroxide	NaOH
lithium hydroxide	LIOH	calcium hydroxide	Ca(OH)2
sulfuric acid	H2504	sulfurous acid	H2503

Section 19 Acids And Bases Answer Key

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Section 19 Acids And Bases Answer Key:

Chemistry Textbook for College and University USA Ibrahim Sikder,2023-06-04 Cehmistry Textbook USA
Chemistry Therald Moeller,2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis This book reviews the chemistry of inorganic substances as the science of matter the units of measure used atoms atomic structure thermochemistry nuclear chemistry molecules and ions in action This text also describes the chemical bonds the representative elements the changes of state water and the hydrosphere which also covers water pollution and water purification Water purification occurs in nature through the usual water cycle and by the action of microorganisms The air flushes dissolved gases and volatile pollutants when water seeps through the soil it filters solids as they settle in the bottom of placid lakes Microorganisms break down large organic molecules containing mostly carbon hydrogen nitrogen oxygen sulfur or phosphorus into harmless molecules and ions This text notes that natural purification occurs if the level of contaminants is not so excessive This textbook is suitable for both chemistry teachers and students

Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose problem solving skills reference and test preparation Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief

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