

8. GAS LAWS AND KINETIC THEORY

Reading:

Hodder +Gibson National 5 Physics:
Bright Red Study Guide:

Pgs. 32-39

Pgs. 30-37

Appropriate websites:

BBC Bitesize: Gas Laws and Kinetic Model

<http://www.bbc.co.uk/education/guides/zjc6fg8/revision>

		green	amber	red
1.	I can use the relationship $P = F \div A$	☐	☐	☐
2.	I know that the temperature of a substance is a measure of the average kinetic energy of its particles.	☐	☐	☐
3.	I can explain the pressure of a gas using the kinetic model.	☐	☐	☐
4.	I can use the Kelvin scale of temperature which describes a temperature of absolute zero, where $E_k = 0$ J.	☐	☐	☐
5.	I know that $0\text{K} = -273^\circ\text{C}$ and can convert from one scale to another.	☐	☐	☐
6.	I can use the relationship $(P_1V_1) = (P_2V_2)$	☐	☐	☐
7.	I can use the relationship $(P_1 / T_1) = (P_2 / T_2)$	☐	☐	☐
8.	I can use the relationship $(V_1 / T_1) = (V_2 / T_2)$	☐	☐	☐
9.	I can explain the above relationships using the kinetic model.	☐	☐	☐

Section 1 Kinetic Theory Study Guide Answers

David D. Reid



Section 1 Kinetic Theory Study Guide Answers:

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