

- 8 The gas in a can of hairspray is made up mainly of a hydrocarbon X. The boiling point of the hydrocarbon X is -5°C and its melting point is -120°C .

(a) Complete the table below by making use of the above information.

temperature ($^{\circ}\text{C}$)	state of hydrocarbon X
30	
-25	liquid
-60	

[1]

- (b) If 'O' represents a molecule of hydrocarbon X, show the arrangement of the molecules at -25°C and at 30°C , in the boxes below.



-25°C



30°C

[2]

- (c) Hydrocarbon X is usually kept as a liquid in aerosol sprays by pressurizing

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