

**Directed Reading for
Content Mastery****Overview
Chemical Bonds**

Directions: All of the statements below are false as written. In the space provided, write a term or phrase that makes the statement true when it is substituted for the underlined words.

- _____ 1. The properties of a compound are the same as the properties of the elements that it contains.
- _____ 2. Superscript numbers in chemical formulas tell how many atoms of each element are found in a unit of compound.
- _____ 3. All the noble gases except helium have 18 electrons in their outer energy level.
- _____ 4. A(n) chemical formula is the force that holds atoms together in a compound.
- _____ 5. An ion is a(n) neutral particle that has either more or fewer electrons than protons.
- _____ 6. Oxidation numbers are written as subscripts.
- _____ 7. A(n) covalent bond is the force of attraction between the opposite charges of the ions in an ionic compound.
- _____ 8. The charge on a compound is always positive.
- _____ 9. Equal sharing of electrons in covalent bonds results in polar molecules.
- _____ 10. Only two identical atoms can share electrons unequally.
- _____ 11. A binary compound contains five different elements.
- _____ 12. An oxidation number tells how many protons an atom must gain, lose, or share to become stable.
- _____ 13. The oxidation number of the copper(II) ion is 3+.
- _____ 14. When writing chemical formulas, add superscripts so that the sum of the oxidation numbers equals ten.
- _____ 15. A polyatomic ion never has a positive or negative charge.
- _____ 16. The polyatomic ion SO_4^{2-} is called the sulfide ion.

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F Rizvi



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Study Guide for The Human Body in Health & Disease - E-Book Kevin T. Patton, Frank B. Bell, Terry

Thompson, Peggie L. Williamson, Eric L. Sun, 2023-03-02 Reinforce your understanding of the concepts in Patton's The Human Body in Health Disease Corresponding to the chapters in the text this study guide reviews essential medical terminology concepts and processes related to anatomy and physiology and explains how body systems function in health and disease Each chapter begins with a quick synopsis of the key points in the textbook chapter A variety of exercises make it easier to review and apply key concepts and labeling of anatomy drawings helps in learning anatomical terms and structures Know your Medical Terms feature helps familiarize you with the various word parts used in medical terminology and pairs nicely with the Language of Medicine word lists in the main text to reinforce medical terminology concepts that are key to understanding A P Brief synopsis of core textbook concepts provides a comprehensive review of essential content Crossword Puzzles and Word Finds help you master new vocabulary terms Application Questions ask you to make judgments based on the information in the chapter Diagrams and labeling exercises help reinforce where the structures of the body are located Matching and fill in the blank exercises aid in understanding chapter content Answers to exercises are provided in the back of the guide along with references to the appropriate textbook page and the textbook objective so that you can easily find where the material is presented Study tips in the Preface provide an overview of the most effective methods for learning and retaining information

, Chemistry 'O' Level Rex M. Heyworth, 2007

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