

World History 7.2 Revolution Brings Reform and Terror (222-227)

Guiding Question: How did the French Revolution change society, and how did it move away from the original goals?

1. Describe the events of the evening of August 4, 1789

On August 4, 1789, a number of noblemen gave rousing addresses in which they proclaimed their love for liberty and equality.

2. What are 3 main points in the Declaration of the Rights of Man?

The three basic tenets of the Declaration of Human Rights are freedom, property rights, and the right to oppose persecution.

3. How does the slogan "Liberty, Equality, Fraternity" sum up the goals of the revolution?

It is the necessity for France to be united, for all citizens to have equal rights and treatment, and for emancipation from previous class structures.

Estates -social classes

4. How did the National Assembly actions affect the Catholic Church?

Church leaders and priests were ordered to be elected and compensated like state officials when the assembly confiscated church properties. As a result, the Catholic Church lost both its political independence and its holdings.

5. Why was the Assembly hesitant to further tax the Bourgeoisie?

France had debt that needed to be paid off, and they were using high taxes and exorbitant costs to accomplish so. They were reluctant to impose additional taxes on the bourgeoisie since they already taxed them excessively and more heavily than other estates, and they feared a third estate uprising against them.

6. Why did the Assembly's actions anger the peasants and how would this affect future revolutionary action?

They defected the estate system because it kept them at the bottom and benefited the wealthy elite. Future revolutions would be impacted by this. It urged that the monarch implement economic reforms to guarantee that the populace had access to food. This group was known as the "3rd estate" and represented France's common people.

7. Why did Louis XIV and his family try to escape France?

Louis XVI reflected on his future as a king. Some of his advisors alerted him to the danger his family and he were in. Many monarchy supporters left France because they believed it to be unsafe. The royal family then attempted to flee France for the Austrian Netherlands in June 1791. However, as they got close to the border, they were stopped and taken back to Paris under supervision.

8. What type of government did the National Assembly create in 1791 and how would power be distributed?

The national assembly established a constitutional monarchy in 1791, with the elected parliament having the authority to pass and implement laws while the king served as the head of state.

9. Summarize the three general groups of the Legislative Assembly: (Radicals, Moderates, and Conservatives)

Radicals are those who want a radical overhaul of the social order or the entire political system (Set on the left) The moderates sought certain governmental changes, but fewer than the extremists (set in the middle) Conservatives supported the notion of a limited monarchy and favored little changes to the current system of governance (set on right)

10. Why did the Legislative Assembly declare war on Austria?

Following Austria's refusal to withdraw its troops from the French border and to retract its alleged threat to use force, When Austria and Prussia responded to France's declaration of war in the spring of 1792 with a concerted invasion, it was ultimately repulsed in the Battle of Valmy in September.

11. Why did Parisians imprison the Royal Family?

The newly established legislative assembly aimed to spread the uprising over all of Europe. France declared war on Austria in April 1792. Louis XVI was blamed for how poorly the war went. He lost his title and was sent in jail after being charged with treason.

12. What were the effects of the war on the daily life in Paris?

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Wolfgang Guggemos

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web may 5 2018 in this video the synthesis of benzoic acid is shown usually a phase transfer catalyst is used but as it can be seen it still works without one patreon h

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web name the given alkene and predict the products of its reaction with i kmno4 in aqueous acid and ii kmno4 in aqueous naoh predict the alkene give the following products on oxidative cleavage with kmno4 in acidic solution draw the product formed when cyclohexene undergoes a reaction with cold dilute basic solution of kmno 4

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web feb 14 2014 we have therefore started our studies by identifying a theory level that properly describes the competition between the pathways a d presented in fig 1 fig 1 possible reactions of toluene and with permanganate anion at positions a c m b c 1 c 2 c c 2 c 3 d c 3 c 4 full size image

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web apr 26 2012 mechanism of arene side chain oxidation by permanganate when treated with hot concentrated acidic kmnox4 k m n o x 4 arenes are oxidised to the corresponding carboxylic acids for example toluene is oxidised to benzoic acid understanding c h bond oxidations h and h transfer in the oxidation - Jul 02 2022

web the oxidation of toluene by permanganate has been studied as a model for the oxidation of c h bonds by metal reagents metalloenzymes and metal oxide surfaces in water the reaction proceeds by hydride h transfer from toluene to a permanganate oxygen whereas in toluene solution permanganate abstracts a hydrogen atom h

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web toluene is oxidized to benzoic acid and a small amount of benzaldehyde the kinetics of the reactions monitored by uv vis spectrometry show that the initial reactions are first order in the concentrations of both nbu4nmno4 and substrate no induction periods are observed

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web an improved kinetic model for the high temperature oxidation of toluene has been developed using previously established reaction mechanisms for benzene and toluene

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web oct 1 2020 in situ drifts combined with ptr tof ms quasi in situ xps and uv vis drs were introduced to learn the process of toluene oxidation confirmed that the reaction mechanism over catalyst followed the mars van krevelen mechanism and surface lattice oxygen played an important role in deeper oxidation toluene

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web mechanism of the oxidation of alcohols with kmno4 ask question asked 7 years 4 months ago modified 7 years 4 months ago viewed 29k times 10 many oxidising agents like chromate dichromate iodine in naoh n a o h etc seem to work via ester formation and elimination

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web may 15 2020 toluene 1 oxidation with kmno4 mechanism 2 benzoic acid $\text{C}_6\text{H}_5\text{COOH}$ SOCl_2 3 benzoyl chloride $\text{C}_6\text{H}_5\text{COCl}$ LiAlH_4 $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$ 4 benzaldehyde $\text{C}_6\text{H}_5\text{CHO}$ $\text{CH}_2\text{OHCH}_2\text{OH}$ h step 1 toluene oxidation with kmno4 kmno4 is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

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web may 1 2004 oxidation of toluene using aqueous potassium permanganate was studied under heterogeneous condition in the presence of hydrodynamic cavitation and compared with the results of the reaction under acoustic cavitation

can acidified or neutral kmno4 oxidise toluene to benzoic acid - Jan 08 2023

web oct 15 2014 1 answer sorted by 6 here are the three equations describing the reduction of manganese and concurrent oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively $\text{MnO}_4^- + 7\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$ $\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$ $\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$ basic $\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$ basic

what is the mechanism of toluene oxidation by kmno4 reddit - Jul 14 2023

web mar 15 2019 what is the mechanism of toluene oxidation by kmno4 it is quite simple reaction that kmno4 can oxidize benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

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web description treatment of an alkylbenzene with potassium permanganate results in oxidation to give the benzoic acid notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a hydrogen attached to the carbon examples

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