Section 9.2 Substituted Hydrocarbons (pages 272-274)

This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.

Reading Strategy (page 272)

Monitoring Your Understanding As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

| Connecting Functional Groups to Types of Compounds | | |
|--|------------------|--|
| Functional Group | Type of Compound | |
| -OH | | |
| -соон | | |
| -NH, | | |

1. Name the two main products when methane and chlorine react.

a. _____

b. _____

3. Describe a substituted hydrocarbon.

 Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound.

Alcohols (page 273)

- Methanol and ethanol are two examples of a class of organic compounds called ______.
- The functional group in an alcohol is represented as -OH and is called a(n) _____ group.
- 7. Identify two ways a halocarbon can be produced.

а.

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p- =

Section 2 Carbon And Hydrocarbons Answers

Amar Bhutani

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